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				enhanced
NEWS	4	APR	07	STN is raising the limits on saved answers
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				information
NEWS	6	APR	26	USPATFULL and USPAT2 enhanced with patent
				assignment/reassignment information
NEWS	7	APR	28	CAS patent authority coverage expanded
NEWS	8	APR	28	ENCOMPLIT/ENCOMPLIT2 search fields enhanced
NEWS	9	APR	28	Limits doubled for structure searching in CAS
				REGISTRY
NEWS	10	MAY	08	STN Express, Version 8.4, now available
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NEWS	12	MAY	11	BEILSTEIN substance information now available on
				STN Easy
NEWS	13	MAY	14	DGENE, PCTGEN and USGENE enhanced with increased
				limits for exact sequence match searches and
				introduction of free HIT display format
NEWS	14	MAY	15	INPADOCDB and INPAFAMDB enhanced with Chinese legal
				status data
NEWS	15	MAY	28	CAS databases on STN enhanced with NANO super role in
				records back to 1992
NEWS	16	JUN	01	CAS REGISTRY Source of Registration (SR) searching
				enhanced on STN
NEWS	17	JUN	26	NUTRACEUT and PHARMAML no longer updated
NEWS	18	JUN	29	IMSCOPROFILE now reloaded monthly
NEWS	19	JUN	29	EPFULL adds Simultaneous Left and Right Truncation
				(SLART) to AB, MCLM, and TI fields
NEWS	20	JUL	09	PATDPAFULL adds Simultaneous Left and Right
				Truncation (SLART) to AB, CLM, MCLM, and TI fields
NEWS	21	JUL	14	USGENE enhances coverage of patent sequence location
				(PSL) data
NEWS	22	JUL	14	CA/CAplus to be enhanced with new citing references
				features
NEWS	23	JUL	16	GBFULL adds patent backfile data to 1855
NEWS	EXP	RESS	MAY	26 09 CURRENT WINDOWS VERSION IS V8.4,
			AND	CURRENT DISCOVER FILE IS DATED 06 APRIL 2009.
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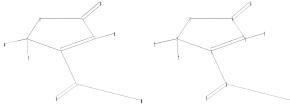
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chain nodes : 6 7 8 9 10 11 12

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ring nodes :
1 2 3 4 5
chain bonds :
1-8 2-11 2-12 4-6 5-7 8-9 8-10
ring bonds :
1-2 1-5 2-3 3-4 4-5
exact/norm bonds :
1-2 1-5 2-3 2-12 3-4 4-5 4-6
exact bonds :
1-8 2-11 5-7
normalized bonds :
8-9 8-10
Match level :
1:Atom 2:Atom 3:Atom 4:Atom 5:Atom 6:CLASS 7:CLASS 8:CLASS 9:CLASS
10:CLASS 11:CLASS 12:CLASS
Generic attributes :
12:
Number of Carbon Atoms : 7 or more
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## L1 STRUCTURE UPLOADED

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Structure attributes must be viewed using STN Express query preparation.

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=> s 11 full
FULL SEARCH INITIATED 20:10:14 FILE 'REGISTRY'
FULL SCREEN SEARCH COMPLETED - 1886 TO ITERATE

100.0% PROCESSED 1886 ITERATIONS 16 ANSWERS
SEARCH TIME: 00.00.01
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FILL ESTIMATED COST

SINCE FILE TOTAL ENTRY SESSION 185.88 186.10

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FILE LAST UPDATED: 19 Jul 2009 (20090719/ED)
REVISED CLASS FIELDS (/NCL) LAST RELOADED: Jun 2009
USPIO MANUAL OF CLASSIFICATIONS THESAURUS ISSUE DATE: Jun 2009

CAplus now includes complete International Patent Classification (IPC) reclassification data for the second quarter of 2009.

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http://www.cas.org/legal/infopolicy.html

This file contains CAS Registry Numbers for easy and accurate substance identification.

The ALL, BIB, MAX, and STD display formats in the CA/CAplus family of databases will soon be updated to include new citing references information. This enhancement may impact record import into database management software. For additional information, refer to NEWS 22.

=> s 12 full L3 53 L2

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L3 ANSWER 1 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2007:840348 CAPLUS

DOCUMENT NUMBER: 147:371328

TITLE: Separation of a mixture of paraconic acids from

Cetraria islandica (L.) Ach. employing a fluorous

tag-catch and release strategy AUTHOR(S):

Horhant, David; Le Lamer, Anne-Cecile; Boustie, Joeel;

Uriac, Philippe; Gouault, Nicolas

CORPORATE SOURCE: UFR Sciences Pharmaceutiques et Biologiques, Universite de Rennes 1, Rennes, 35043, Fr.

SOURCE: Tetrahedron Letters (2007), 48(34), 6031-6033

CODEN: TELEAY; ISSN: 0040-4039

PUBLISHER: Elsevier Ltd. DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 147:371328

AB A light-fluorous catch and release approach application has been designed to the separation of a mixture of three paraconic acids extracted from the Island

moss (Cetraria islandica (L.) Ach.). The (+)-protolichesterinic acid was caught and released via a Michael/retro-Michael addition sequence with a fluorous thiol, while the resulting two other compds, were classically separated, allowing the isolation of (+)-roccellaric acid for the first time in this lichen.

70579-62-3P, (+)-Lichesterinic acid

RL: BSU (Biological study, unclassified); PUR (Purification or recovery);

BIOL (Biological study); PREP (Preparation) (separation of a mixture of paraconic acids from Cetraria islandica

employing

a fluorous tag-catch and release strategy) RM 70579-62-3 CAPLUS

3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2R)- (CA INDEX NAME)

Absolute stereochemistry.

REFERENCE COUNT:

48 THERE ARE 48 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT L3 ANSWER 2 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2004:60242 CAPLUS

DOCUMENT NUMBER: 140:111267

TITLE: Preparation of  $\gamma$ -butyrolactone-4-carboxylate derivatives as inhibitors of fatty acid synthase

INVENTOR(S): Kuhadja, Francis P.; Medghalchi, Susan M.; Thupari, Jagan N.; Townsend, Craig A.; McFadden, Jill M.

PATENT ASSIGNEE(S): Fasgen, Llc., USA; The Johns Hopkins University

SOURCE: PCT Int. Appl., 57 pp.

CODEN: PIXXD2
DOCUMENT TYPE: Patent

LANGUAGE: English FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PA:	ENT :	NO.			KIN	D	DATE				LICAT				D	ATE	
	WO 2004006835 WO 2004006835									WO 2003-US20960								
		W:	AE,	AG.	AL.	AM.	AT.	AU.	AZ.	BA.	BE	, BG,	BR.	BY.	BZ.	CA.	CH.	CN.
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			GM.	HR.	HU,	ID.	IL.	IN.	IS.	JP.	KE	, KG,	KP.	KR.	KZ.	LC.	LK.	LR.
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			FI.	FR.	GB,	GR,	HU.	IE.	IT.	LU,	MC	, NL,	PT.	RO,	SE,	SI,	SK,	TR.
			BF,	ВJ,	CF,	CG,	CI,	CM,	GA,	GN,	GÇ	, GW,	ML,	MR,	NE,	SN,	TD,	TG
	CA	2491	183			A1		2004	0122		CA	2003-	2491	183		2	0030	701
	AU	2003	2488	10		A1		2004	0202		AU	2003-	2488	10		2	0030	701
	EΡ	1534	263			A2		2005	0601		EP	2003-	7643	43		2	0030	701
		R:	AT,	BE,	CH,	DE,	DK,	ES,	FR,	GB,	GF	, IT,	LI,	LU,	NL,	SE,	MC,	PT,
			IE,	SI,	LT,	LV,	FI,	RO,	MK,	CY,	AL	, TR,	BG,	CZ,	EE,	HU,	SK	
	JP 2005533107					T		2005	1104		JP	2004-	5215	21		2	0030	701
		1705				A		2005	1207		CN	2003-	8183	69		2	0030	701
	ΙN	2004	KN02	001		A		2007	0309		IN	2004-	KN20	01		2	0041	229
	US	2006	0241	177		A1		2006	1026		US	2006-	5198	04		2	0060	519
	IN 2008KN02395					Α		2009	0123		IN	2008-	KN23	95		2	0080	613
PRIOR	PRIORITY APPLN. INFO.:										US	2002-	3928	09P		P 2	0020	701
											WO	2003-	US20	960		W 2	0030	701
											IN	2004-	KN20	01		A3 2	0041	229
OTHER	0.0	TIDOR	101.			142 D	200	1 10.	1110	c -								

OTHER SOURCE(S): MARPAT 140:111267

GI

AB The title compds. I [R1 = H, (cyclo)alkyl, alkenyl, (alkyl)aryl, etc.; R2 = (cyclo)alkyl, alkenyl, (alkyl)aryl, etc.; X = OR3 or NHR3, where R3 = H, (cyclo)alkyl, alkenyl, (alkyl)aryl, etc.] were prepared as inhibitors of fatty acid synthase and neuropeptide-Y for weight loss, anti-microbial and anti-cancer applications. Thus, reaction of

 $(\pm)-\alpha-methylene-\gamma-butyrolactone-5-hexyl-4-carboxylic acid with allylamine yielded compound II. The latter inhibits human fatty acid$ 

ΙI

synthase with IC50 = 81  $\mu$ g/mL. IT 647830-53-3P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation of  $\gamma$ -butyrolactone carboxylate derivs. as inhibitors of fatty acid synthase)

RN 647830-53-3 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-2-octyl-5-oxo- (CA INDEX NAME)

REFERENCE COUNT:

Me

THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT L3 ANSWER 3 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2003:4431 CAPLUS

DOCUMENT NUMBER: 138:254998

TITLE: Vicinal diamion of triethyl ethanetricarboxylate: syntheses of (±)-lichesterinic acid,

(±)-phaseolinic acid, (±)-nephromopsinic acid,

(±)-rocellaric acid, and

(±)-dihydroprotolichesterinic acid

AUTHOR(S): Pohmakotr, Manat; Harnying, Wacharee; Tuchinda, Patoomratana; Reutrakul, Vichai

CORPORATE SOURCE: Department of Chemistry, Faculty of Science, Mahidol

University, Bangkok, 10400, Thailand

SOURCE: Helvetica Chimica Acta (2002), 85(11), 3792-3813 CODEN: HCACAV; ISSN: 0018-019X

PUBLISHER: Verlag Helvetica Chimica Acta

DOCUMENT TYPE: Journal

LANGUAGE: English

OTHER SOURCE(S): CASREACT 138:254998

GI

- AB The vicinal diamion derived from tri-Et ethanetricarboxylate reacted regioselectively with aldehydes and ketones at C(B) to provide paraconic acid derivs. I [R = 4-MeOC6H4, Me3C, Me(CH2)4, etc.] in moderate to high yields as mixts. of diastereoisomers. The paraconic acid derivs. II [R = Me(CH2)1, n = 4, 12] were utilized as the starting materials for the syntheses of (t)-lichesterinic acid, (t)-phaseolinic acid, (t)-nephromopopinic acid, (t)-rocellalric acid, and
  - (±)-dihydroprotolichesterinic acid.

493-47-0P, (±)-Lichesterinic acid
RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of (±)-lichesterinic acid, (±)-phaseolinic acid,

(±)-nephromopsinic acid, (±)-rocellaric acid, and

(±)-dihydroprotolichesterinic acid from γ-lactones derived

from lactonization of carbonyl compds. with tri-Et

ethanetricarboxylate)

- RN 493-47-0 CAPLUS
- CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX NAME)

REFERENCE COUNT:

51 THERE ARE 51 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 4 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1999:665856 CAPLUS

DOCUMENT NUMBER: 132:33194

TITLE: A Revised Structure for (-)-Dihydropertusaric Acid, a  $\gamma$ -Butyrolactone Acid from the Lichen Punctelia

microsticta

AUTHOR(S): Maier, Marta S.; Gonzalez Marimon, Diego I.; Stortz,

Carlos A.; Adler, Monica T.

CORPORATE SOURCE: Departamento de Quimica Organica and Departamento de Ciencias Biologicas, Facultad de Ciencias Exactas v

Naturales, Buenos Aires, 1428, Argent.

SOURCE: Journal of Natural Products (1999), 62(11), 1565-1567

CODEN: JNPRDF; ISSN: 0163-3864

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

GI

$$H_{3}C-CO+CH_{2}+CH_{2}$$
  $CH_{2}$   $CH$ 

- AB The y-butyrolactone acid, (-)-dihydropertusaric acid (I), and two known compds., (-)-isomuronic acid and the tridepside gyrophoric acid, were isolated from the lichen Punctelia microsticta. The structure and stereochem. of I were determined on the basis of spectroscopic evidence and mol. modeling. Spectroscopic and phys. data of I were identical with those of a previously isolated compound from the lichen Pertusaria albescens which had been reported with a different relative configuration.
- II 70579-66-7P, (-)-Isomuronic acid Rl: BOC (Biological occurrence); BSU (Biological study, unclassified); PUR (Purification or recovery); BIOL (Biological study); OCCU (Occurrence); PREP (Preparation)
  - (isolation, mol. structure, conformation, and revised configuration for (-)-dihydropertusaric acid, a  $\gamma$ -butyrolactone acid from the lichen Punctelia microstictal

RN 70579-66-7 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-(14-oxopentadecyl)-, (2R)- (CA INDEX NAME)

Absolute stereochemistry.

REFERENCE COUNT:

18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 5 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1999:561920 CAPLUS

DOCUMENT NUMBER: 131:226128

TITLE: Some lichen products have antimicrobial activity

AUTHOR(S): Garcia Rowe, J.; Garcia Gimenez, M. D.; Saenz

Rodriguez, M. T.

CORPORATE SOURCE: Lab. Vegetal Biology, Faculty Pharmacy, Univ. Seville,

Seville, Spain

SOURCE: Zeitschrift fuer Naturforschung, C: Journal of Biosciences (1999), 54(7/8), 605-609

BIOSCIENCES (1999), 34(7/0), 603-609

CODEN: ZNCBDA; ISSN: 0341-0382

PUBLISHER: Verlag der Zeitschrift fuer Naturforschung

DOCUMENT TYPE: Journal

LANGUAGE: English

B Antimicrobial activity in some lichens from south Spain was studied. Some lichenical substances are also identified. The structures of all compds. were elucidated by phys., spectral and chemical methods. A very high activity against Gram-pos. bacteria was observed in lichens containing usnic acid.

I 493-47-0P, Lichesteric acid

RL: BAC (Biological activity or effector, except adverse); BOC (Biological occurrence); BSU (Biological study, unclassified); PRP (Properties); PUR (Purification or recovery); BIOL (Biological study); OCCU (Occurrence); PREP (Preparation)

(lichen products with antimicrobial activity)

N 493-47-0 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX NAME)

REFERENCE COUNT:

12 THERE ARE 12 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L3 ANSWER 6 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1995:834162 CAPLUS

DOCUMENT NUMBER: 123:275351

ORIGINAL REFERENCE NO.: 123:48943a,48946a
TITLE: Screening of tissue

TITLE: Screening of tissue cultures and thalli of lichens and some of their active constituents for inhibition of

tumor promoter-induced Epstein-Barr virus activation
Yamamoto, Yoshikazu, Miura, Yasutaka; Kinoshita,
Yasuhiro; Hiouchi, Masako; Yamada, Yasuvuki; Murakami,

Akira; Ohigashi, Hajime; Koshimizu, Koichi
CORPORATE SOURCE: Central Res. Inst., Nippon Paint Co., Ltd., Osaka,

572, Japan
SOURCE: Chemical & Pharmaceutical Bulletin (1995), 43(8),

1388-90 CODEN: CPBTAL; ISSN: 0009-2363

PUBLISHER: Pharmaceutical Society of Japan

DOCUMENT TYPE: Journal

LANGUAGE: English

AB Inhibition of tumor promoter-induced Epstein-Barr virus (EBV) activation was screened using tissue culture and thallus exts. of lichens. Usnea longissima ACH. thallus and Cetraria ornate Muell. Arg. tissue culture showed strong inhibitor activity. The authors identified (+)-usnic acid (1), barbatic acid (2), diffractaic acid (3), 4-0-demethylbarbatic acid (4), and evernic acid (5) as inhibitors of EBV activation from the U. longissima thallus. Of these compds., (+)-usnic acid exhibited the highest inhibitory activity (ICSO = 1.0 mM).

IT 493-47-0, Lichesterinic acid

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); BIOL (Biological study)

(screening in tissue cultures and thalli of lichens and some of their active constituents for inhibition of tumor promoter-induced Epstein-Barr virus activation)

RN 493-47-0 CAPLUS

NAME)

L3 ANSWER 7 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1995:571012 CAPLUS

DOCUMENT NUMBER: 122:306540

ORIGINAL REFERENCE NO.: 122:55533a,55536a

TITLE: Inhibitor of epstein-barr virus expression comprising usnic acid and lichesterinic acid derivatives

INVENTOR(S): Yamamoto, Yoshikazu; Miura, Yasutaka; Kinoshita, Yasuhiro; Ohigashi, Hajime; Koshimizu, Koichi

PATENT ASSIGNEE(S): Nippon Paint Co., Ltd., Japan

SOURCE: Eur. Pat. Appl., 6 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE		
EP 646373	A2	19950405	EP 1994-113368	19940826		
EP 646373	A3	19950726				
R: DE, FR, GB						
JP 07112931	A	19950502	JP 1994-201881	19940826		
PRIORITY APPLN. INFO.:			JP 1993-212632 A	19930827		
			JP 1993-212673 A	19930827		

OTHER SOURCE(S): MARPAT 122:306540

AB Inhibitors of epstein-barr virus expression comprise usnic acid and lichesterinic acid derivs. (Markush structure given). Epstein-barr virus in human lymphoid Raji cells were inhibited by usnic acid (5x10-5) at the rate of 99%.

IT 493-47-0, Lichesterinic acid

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); BIOL (Biological study) (inhibitor of epstein-barr virus expression)

RN 493-47-0 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX NAME)

L3 ANSWER 8 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1995:423858 CAPLUS

DOCUMENT NUMBER: 122:255757

ORIGINAL REFERENCE NO.: 122:46377a,46380a

TITLE: In vitro inhibition of 5-lipoxygenase by protolichesterinic acid from Cetraria islandica

AUTHOR(S): Ingolfsdottir, K.; Breu, W.; Huneck, S.;

Gudjonsdottir, G. A.; Mueller-Jakic, B.; Wagner, H.

CORPORATE SOURCE: Dept. of Pharmacy, University of Iceland, Revkjavik,

101, Iceland

SOURCE: Phytomedicine (1994), 1(3), 187-91 CODEN: PYTOEY; ISSN: 0944-7113

DOCUMENT TYPE: Journal

LANGUAGE: Journal Language: English

AB The aliphatic  $\alpha$ -methylene- $\gamma$ -lactone (+)-protolichesterinic acid,

isolated from Cetraria islandica, has been shown to exhibit inhibitory effects on the enzyme 5-lipoxygenase in an in vitro assay in which porcine leukocytes are used as a source of the enzyme system. The isomeric compds. (+)-lichesterinic acid and (-)-lichesterinic acid, prepared from (+)-protolichesterinic and (-)-allo-protolichesterinic acids, resp., exhibited anti-5-lipoxygenase activity of the same order of magnitude. (+)-Me lichesterinate, however, was inactive. It was shown that despite its lipophilic nature, protolichesterinic acid is extractable into an aqueous

medium, the concentration being dependent on the length of extraction IT 22800-25-5P, (-)-Lichesterinic acid 70579-62-3P,

(+)-Lichesterinic acid

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); SPN (Synthetic preparation); BIOL (Biological study); PREP (Preparation)

(in vitro inhibition of lipoxygenase by protolichesterinic acid from Cetraria islandica)

RN 22800-25-5 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2S)- (CA INDEX NAME)

Absolute stereochemistry.

RN 70579-62-3 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2R)- (CA INDEX NAME)

L3 ANSWER 9 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1994:567 CAPLUS DOCUMENT NUMBER: 120:567

ORIGINAL REFERENCE NO.: 120:135a,138a

TITLE: Acne-controlling antibacterial agents containing usnic

acids or lichesterinic acids

INVENTOR(S): Higuchi, Masako; Miura, Yasutaka; Kinoshita, Yasuhiro;

Yamamoto, Yoshikazu; Mayama, Shiqeyuki

PATENT ASSIGNEE(S): Nippon Paint Co Ltd, Japan

SOURCE: Jpn. Kokai Tokkvo Koho, 3 pp.

CODEN: JKXXAF
DOCUMENT TYPE: Patent

LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 05246822	A	19930924	JP 1992-84686	19920307
RIORITY APPLN. INFO.:			JP 1992-84686	19920307
B Antibacterial agents	arains	t Propingiba	acterium acnes co	ontain usnic acids

AB Antibacterial agents against Propionibacterium acnes contain usnic acids or lichesterinic acids as active ingredients. Lichesterinic acid, protolichesterinic acid, and usnic acid inhibited the growth of P. acnes in vitro.

IT 493-47-0, Lichesterinic acid

RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(antibacterial activity of, against Propionibacterium acnes, for acne treatment)

RN 493-47-0 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX NAME)

L3 ANSWER 10 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1993:445255 CAPLUS

DOCUMENT NUMBER: 119:45255 ORIGINAL REFERENCE NO.: 119:8151a,8154a

TITLE: Studies on Chilean lichens. XVII. Metabolites of

Cetraria chlorophylla

AUTHOR(S): Garbarino, Juan A.; Quilhot, Wanda; Piovano, Marisa;

Figueroa, Yasmin; Torres, Pamela

Dep. Quim., Univ. T. F. Santa Maria, Valparaiso, Chile CORPORATE SOURCE: SOURCE: Revista Latinoamericana de Ouimica (1991), 22(3), 53-4

CODEN: RLAOA8; ISSN: 0370-5943

DOCUMENT TYPE: Journal

LANGUAGE: Spanish Lichesterinic acid, atranorin, and peroxyergosterol were isolated from C.

chlorophylla, a lichen from Continental Chile. The latter compound is reported for the first time for the Cetraria genus.

22800-25-5 IT

RL: BIOL (Biological study)

(of Cetraria chlorophylla from Chile)

RN 22800-25-5 CAPLUS

3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2S)- (CA CN

INDEX NAME)

L3 ANSWER 11 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1993:424317 CAPLUS

DOCUMENT NUMBER: 119:24317

ORIGINAL REFERENCE NO.: 119:4432h,4433a

TITLE: Chemical examination of South Indian lichens: Lobaria

japonica (Zahlbr) Asah and Heterodermia leucomela Borri (Fee') Swinsc & Kroq

Ramesh, P.; Baig, E. Shere Ali

CORPORATE SOURCE: Dep. Nat. Prod. Chem., Kamarai Univ., Madurai, 625

021, India
SOURCE: Indian Journal of Heterocyclic Chemistry (1993), 2(3),

147-8

CODEN: IJCHEI; ISSN: 0971-1627

DOCUMENT TYPE: Journal

LANGUAGE: English

AB From the South Indian iichens L. japonica and H. leucomela, atranorin, salazinic acid, zeorin, (+)-lichesterinic acid, and lecanoric acid were isolated.

IT 70579-62-3, (+)-Lichesterinic acid RL: BIOL (Biological study)

(of lichens, of India)

RN 70579-62-3 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2R)- (CA INDEX NAME)

L3 ANSWER 12 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1993:260713 CAPLUS DOCUMENT NUMBER: 118:260713

ORIGINAL REFERENCE NO.: 118:45203a,45206a

TITLE: Topical preparations containing lichesteric acid INVENTOR(S): Koiso, Ichiro; Matsugami, Michio; Katagiri, Takayuki;

Yokoyama, Koji; Oonuki, Keiko; Nakano, Hiroyuki

PATENT ASSIGNEE(S): Pola Kasei Kogyo Kk, Japan

SOURCE: Jpn. Kokai Tokkvo Koho, 7 pp.

CODEN: JKXXAF

DOCUMENT TYPE: Patent

LANGUAGE: Japanese

FAMILY ACC. NUM. COUNT: 1 PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
JP 05058872	A	19930309	JP 1991-247071	19910830
PRIORITY APPLN. INFO.:			JP 1991-247071	19910830
			ain lichesteric acid (I	
inhibited melanin	formatio	on in B-16 :	melanoma cells by 50.3%	. A skin cream
containing I was :	formulate	ed.		

493-47-0, Lichesteric acid RL: BIOL (Biological study)

(skin-lightening cosmetics containing, melanin formation-inhibiting)

RN 493-47-0 CAPLUS CN

3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX NAME)

L3 ANSWER 13 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1992:630101 CAPLUS

DOCUMENT NUMBER: 117:230101

ORIGINAL REFERENCE NO.: 117:39701a,39704a

TITLE: Contribution to the chemistry of proto- and

allo-protolichesterinic acids

AUTHOR(S): Huneck, Siegfried; Takeda, Reiji

Inst. Pflanzenbiochem., Halle/Saale, D-O-4050, Germany SOURCE:

Zeitschrift fuer Naturforschung, B: Chemical Sciences (1992), 47(6), 842-54

CODEN: ZNBSEN; ISSN: 0932-0776

Journal

DOCUMENT TYPE: LANGUAGE: German GT

The isolation and spectroscopic characterization of (-)-allo-protoichesterinic acid (I) from Cetraria komarovii is described.

Protolichesterinic acid (II) and I were transformed into numerous nitrogen-containing derivs. and the isomerization of the dihydro acids was investigated.

22800-25-5, (-)-Lichesterinic acid RL: BIOL (Biological study)

(of Cetraria komarovii) 22800-25-5 CAPLUS RN

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2S)- (CA INDEX NAME)

Absolute stereochemistry.

70579-62-3P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and chemical transformation reactions of)

70579-62-3 CAPLUS RN

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2R)- (CA INDEX NAME)

L3 ANSWER 14 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1992:462532 CAPLUS DOCUMENT NUMBER: 117:62532

ORIGINAL REFERENCE NO.: 117:10794h,10795a

TITLE: Inhibitory effects of plant secondary metabolites on cytotoxic activity of polymorphonuclear leukocytes

AUTHOR(S): Kinoshita, Kaoru; Morikawa, Kaoru; Fujita, Masahiko;

Natori, Shinsaku

CORPORATE SOURCE: Meiji Coll. Pharm., Tanashi, 188, Japan SOURCE: Planta Medica (1992), 58(2), 137-45

CODEN: PLMEAA; ISSN: 0032-0943

DOCUMENT TYPE: Journal

LANGUAGE: English

The inhibitory effects of 151 natural products, representing most of the frequently occurring types, on the cytotoxicity towards NM2 tumor cells of polymorphonuclear leukocytes (FNN) induced by TAK, a polysaccharide immunomodulator, were examined Forty-two compds. inhibited the TAK-induced activation of FNN. Among them some naturally occurring quinones and various alkaloids (nicotine, Cinchona alkaloids, isoquinoline alkaloids such as cepharanthine, and indole alkaloids such as ajmaline) exhibited potent inhibitory effects. Using the inhibition assay for monitoring the exts. of Hydrangea Dulcis folium, Scopoliae rhizoma, Cinchona cortex, Magnoliae cortex, Stephania tuber, and Rauwolfia radix were analyzed to characterize the active constituents.

IT 493-47-0, Lichesterinic acid RL: BIOL (Biological study)

(cytotoxic activity of polymorphonuclear leukocytes toward neoplasm response to)

response to) RN 493-47-0 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX NAME)

L3 ANSWER 15 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1991:489130 CAPLUS

DOCUMENT NUMBER: 115:89130

ORIGINAL REFERENCE NO.: 115:15247a,15250a

TITLE: The chemical constituents of four lichens from China

Li, Bo; Lin, Zhongwen; Sun, Handong AUTHOR(S):

CORPORATE SOURCE: Kunming Inst. Bot., Acad. Sin., Kunming, 560204, Peop.

Rep. China

Yunnan Zhiwu Yanjiu (1991), 13(1), 81-4

SOURCE: CODEN: YCWCDP: ISSN: 0253-2700

DOCUMENT TYPE: Journal Chinese

LANGUAGE:

The following 18 compds. were isolated and identified from four lichens in China: Me 5-methyl-β-orcinolcarboxylate, orsellinic acid, everninic acid, Me orsellinate, pseudocyphellarin A and lecanoric acid from Sticta henryana Mull. Arag.; atranorin, lecanoric acid, stictic acid, norstictic acid, salazinic acid, fumarprotecetraric acid and (+)-usnic acid from Alectoria variabilis Brystrek; (-)-usnic acid, (-)-lichesterinic acid, (+)-protolichesterinic acid and friedelin from Nephromopis strachyi Mull Arg. ectocarpisma Hue; and Et hematommate and Me β-orcinolcarboxylate from Stereocaulon pomiferum Duvign. The anal. showed that N. strachvi f. ectocarpisma is very rich in antibiotic constituents, such as usnic acid and y-lactonic acids, and that S. pomiferum can be used in producing lichen perfume.

22800-25-5, (-)-Lichesterinic acid

RL: PROC (Process)

(isolation of, from lichen)

RN 22800-25-5 CAPLUS

CN 3-Furancarboxvlic acid, 2,5-dihvdro-4-methvl-5-oxo-2-tridecvl-, (2S)- (CA INDEX NAME)

L3 ANSWER 16 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1991:404422 CAPLUS DOCUMENT NUMBER: 115:4422

ORIGINAL REFERENCE NO.: 115:875a,878a

TITLE: High-performance liquid chromatographic method for the quantitative determination of some organic acids in

lichens

AUTHOR(S): Zhou, Xinru; Kang, Xiaoyu; Ke, Yikan; Yuan, Hancheng;

Da, Jun; Gao, Xianggun

CORPORATE SOURCE: Dep. Appl. Chem., Beijing Inst. Chem. Technol.,

100029, Peop. Rep. China SOURCE: Sepu (1991), 9(2), 128-30

CODEN: SEPUER; ISSN: 1000-8713

DOCUMENT TYPE: Journal

LANGUAGE: Chinese

AB A HPLC method was developed for the determination of usnic acid, lichesterinic acid, and protolichesterinic acid in Cetraria lichens. Conditions for preparing standard reagents for quant. anal. by HPLC were developed as were

methods for extracting usnic acid from lichen samples.

IT 493-47-0, Lichesterinic acid

RL: ANT (Analyte); ANST (Analytical study)
(determination of, in Cetraria lichens by HPLC)

RN 493-47-0 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX

L3 ANSWER 17 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1990:608340 CAPLUS

DOCUMENT NUMBER: 113:208340

ORIGINAL REFERENCE NO.: 113:35121a,35124a

TITLE: Two new aliphatic acids from the lichen Parmotrema praesorediosum

AUTHOR(S): David, Feeya; Elix, John A.; Wahid bin Samsudin, M. CORPORATE SOURCE: Fac. Sci., Prince Songkla Univ., Hat Yai, 90112,

Thailand

SOURCE: Australian Journal of Chemistry (1990), 43(7),

1297-300 CODEN: AJCHAS; ISSN: 0004-9425

DOCUMENT TYPE: Journal

LANGUAGE: English

AB The new aliphatic acids, (+)-praesorediosic acid

[2-(14'-carboxytetradecyl)-4-methyl-5-oxo-2,5-dihydrofuran-3-carboxylic acid] (I) and (+)-protopraesorediosic acid

[2-(14'-carboxytetradecyl)-4-methylene-5-oxo-2,5-tetrahydrofuran-3-carboxylic acid] (II) have been isolated from the lichen P. præsorediosum.

IT 130342-70-0, (+)-Praesorediosic acid

RL: BIOL (Biological study)
(from Parmotrema praesorediosum, isolation and structure of)

RN 130342-70-0 CAPLUS

CN 2-Furanpentadecanoic acid, 3-carboxy-2,5-dihydro-4-methyl-5-oxo-, (2R)-(CA INDEX NAME)

L3 ANSWER 18 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1986:417935 CAPLUS

DOCUMENT NUMBER: 105:17935 ORIGINAL REFERENCE NO.: 105:2857a

TITLE: Effect of lichesterinic acid and sarkomycin on the

permeability of biological membranes

AUTHOR(S): Omarov, I. A.; Gaibov, T. D.; Akhmedov, G. I.

CORPORATE SOURCE: Azerb. Gos. Univ., Baku, USSR

SOURCE: Izvestiva Akademii Nauk Azerbaidzhanskoi SSR, Seriva

Biologicheskikh Nauk (1986), (1), 106-12

CODEN: IABLAQ; ISSN: 0132-6112

DOCUMENT TYPE: Journal

LANGUAGE: Russian
AB Lichesterinic acid (I) [493-47-0] (5 mg/kg for 10 days) and the

Lichesterinic acid (I) [493-47-0] (5 mg/kg for 10 days) and the antitumor agent sarkomycin (II) [11031-48-4] (4 mg/kg for 12 days) increased both cellular (erythrocyte) and vascular permeability to indicator substances in rats. The effects were reversible, and were greatly diminished 10 days after cessation of drug administration. The changes induced by I were less marked than those induced by II. Both I and II induced marked changes in the Na+ and K+ content of erythrocytes.

IT 493-47-0 RL: BIOL (Biological study)

(cellular and vascular permeability enhancement by)

RN 493-47-0 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX NAME)

L3 ANSWER 19 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1986:183270 CAPLUS

DOCUMENT NUMBER: 104:183270

ORIGINAL REFERENCE NO.: 104:28969a,28972a

TITLE: Lichen substances. Part 144. (-)-Allo-pertusaric acid and (-)-dihydropertusaric acid from the lichen

Pertusaria albescens

AUTHOR(S): Huneck, Siegfried; Toensberg, Tor; Bohlmann, Ferdinand

CORPORATE SOURCE: Inst. Plant Biochem., Ger. Acad. Sci., Halle/Saale,

4010, Ger. Dem. Rep.

SOURCE: Phytochemistry (Elsevier) (1986), 25(2), 453-9

CODEN: PYTCAS; ISSN: 0031-9422

DOCUMENT TYPE: Journal LANGUAGE: English

LANGUAGE:

AB The structures of 2 y-lactone carboxylic acids from the lichen P. albescens, (-)-allo-pertusaric acid (II) and (-)-dihydropertusaric acid (III), were elucidated by spectroscopic and chemical methods. From P. ophthalmiza, taraxerone and a mixture of long-chain aliphatic alcs. and fatty acids were isolated.

IT 72960-05-5P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(preparation and reactions of)

RN 72960-05-5 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-(14-oxopentadecyl)-, (S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry.

IT 101899-71-2P

RL: SPN (Synthetic preparation); PREP (Preparation)
(preparation of)

RN 101899-71-2 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-2-[14-(hydroxyimino)pentadecyl]-4-methyl-5-oxo-, (S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Double bond geometry unknown.

L3 ANSWER 20 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1986:182651 CAPLUS

DOCUMENT NUMBER: 104:182651

ORIGINAL REFERENCE NO.: 104:28861a,28864a

TITLE: A high performance liquid chromatographic method for

the analysis of lichen compounds from the genera

Cladina and Cladonia

Huovinen, K.; Hiltunen, R.; Von Schantz, M.

CORPORATE SOURCE: Sch. Pharm., Univ. Helsinki, Helsinki, SF-00170, Finland

SOURCE: Acta Pharmaceutica Fennica (1985), 94(3), 99-112 CODEN: APHFDO; ISSN: 0356-3456

DOCUMENT TYPE: Journal

LANGUAGE: English

Reversed-phase HPLC for determination of aromatic lichen acids in Cladina and Cladonia species was done on a 250 + 4-mm inner diameter column packed with LiChrosorb RP-8, 5-µm, fitted with a 30 + 4-mm inner diameter Precolumn packed with Perisorb RP-8, 30-40-µm, with a mobile phase elution gradient of MeOH in H2O. The lichen acids were extracted with Me2CO-EtOH-DMF (40:40:20), and benzoic acid and bis(2-hexylethyl) phthalate were used as internal stds. compds. Identities were confirmed by TLC on silica gel. UV detection at 270-nm and 254 nm was used. Retention indexes were determined for the compds. and their reproducibility ranged 0.09-0.56%. Intra-assay relative standard deviation ranged 2.1-5.5%

and inter-assay relative standard deviation ranged 3.1-14.9%. The method may be useful in chemotaxonomic studies of lichens, with sensitivity of the

technique making micropopulation studies possible.

493-47-0

RL: ANT (Analyte); ANST (Analytical study)

(determination of, in lichens by reversed-phase HPLC with UV detection) RM 493-47-0 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX NAME)

L3 ANSWER 21 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1983:403006 CAPLUS DOCUMENT NUMBER: 99:3006

ORIGINAL REFERENCE NO.: 99:595a,598a

TITLE: Structural elucidation of 13-acetoxylichesterinic and 13-acetoxyprotolichesterinic acids, two aliphatic lichen metabolites from Neuropogon trachycarpus

Ghogomu, Raphael Tih; Bodo, Bernard

CORPORATE SOURCE: Lab. Chim. Appl. Org., Mus. Natl. Hist. Nat., Paris, 75005, Fr.

SOURCE: Phytochemistry (Elsevier) (1982), 21(9), 2355-8

CODEN: PYTCAS; ISSN: 0031-9422

DOCUMENT TYPE: Journal English

LANGUAGE:

Examination of the lichen N. trachycarpus yielded 6 aliphatic acids related to lichesterinic acid, neuropogolic, murolic, isomuronic, and muronic acids, and 2 new compds., 13-acetoxylichesterinic and

13-acetoxyprotolichesterinic acids (I and II resp.), the structures of

which were determined by chemical and spectral means.

70579-66-7 75716-00-6 RL: BIOL (Biological study)

(from Neuropogon trachycarpus)

RN 70579-66-7 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-(14-oxopentadecyl)-, (2R)- (CA INDEX NAME)

Absolute stereochemistry.

RN 75716-00-6 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-2-(14-hydroxypentadecyl)-4-methyl-5oxo- (9CI) (CA INDEX NAME)

- IT 85644-00-4
  RI: BIOL (Biological study)
  (from Neuropogon trachycarpus, structure of)
- RN 85644-00-4 CAPLUS
- CN 3-Furancarboxylic acid, 2-[13-(acetyloxy)tridecyl]-2,5-dihydro-4-methyl-5-oxo-, (R)- (9CI) (CA INDEX NAME)

L3 ANSWER 22 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1982:420105 CAPLUS

DOCUMENT NUMBER: 97:20105 ORIGINAL REFERENCE NO.: 97:3505a,3508a

TITLE: Substitution of methyl tert-butyl ether for diethyl

ether in the standardized thin-layer-chromatographic

method for lichen products Culberson, C. F.; Johnson, A.

Dep. Bot., Duke Univ., Durham, NC, 27706, USA CORPORATE SOURCE:

Journal of Chromatography (1982), 238(2), 483-7

CODEN: JOCRAM: ISSN: 0021-9673

DOCUMENT TYPE: Journal

SOURCE:

LANGUAGE: English

In the common 3-developer thin-layer-chromatog. (TLC) method for the identification of lichen products, solvent system B was modified by substituting Me tert-Bu ether for Et20 because of problems of evaporation and storage of Et20. Modified solvent B, which contains hexane-Me tert-Bu ether-HCO2H (140:72:18), has chromatog, properties nearly identical to those of unmodified solvent B, which contains hexane-Et20-HCO2H (120:90:20). TLC was done on 12.5-cm-long Merck silica gel 60 F254 plates with atranorin and norstictic acid as internal controls. Standardized Rf data for modified solvent B are given for all major classes of lichen products. Me tert-Bu ether also is recommended for use as extraction solvent in the procedure for the hydrolysis of lichen depsides.

493-47-0 RL: ANT (Analyte); ANST (Analytical study)

(chromatog. of, thin-layer, of lichens, solvent for)

RN 493-47-0 CAPLUS

CN 3-Furancarboxvlic acid, 2,5-dihvdro-4-methvl-5-oxo-2-tridecvl- (CA INDEX NAME)

L3 ANSWER 23 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1982:214247 CAPLUS

DOCUMENT NUMBER: 96:214247 96:35336h,35337a ORIGINAL REFERENCE NO .:

TITLE: Quinones of the lichen Cetraria cucullata

AUTHOR(S): Krivoshchekova, O. E.; Maximov, O. B.; Stepanenko, L.

S.; Mishchenko, N. P.

Pacific Inst. Bioorg. Chem., Far East Sci. Cent., Vladivostok, 22, USSR

SOURCE:

Phytochemistry (Elsevier) (1982), 21(1), 193-6

CODEN: PYTCAS; ISSN: 0031-9422

DOCUMENT TYPE: Journal

LANGUAGE: English

In addition to known compds., the monomeric and dimeric quinones I (n = 0, 1) AB were isolated from C. cucullata, and their structures determined by chemical and

I

spectral methods. A third pigment was isolated in small amts. but its structure was not determined

70579-62-3

RL: BIOL (Biological study)

(from Cetraria cucullata)

RN 70579-62-3 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2R)- (CA INDEX NAME)

L3 ANSWER 24 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1980:635083 CAPLUS

DOCUMENT NUMBER: 93:235083

ORIGINAL REFERENCE NO.: 93:37598h,37599a

TITLE: Structure of isomuronic and neuropogolic acids, new

aliphatic acids from the lichen, Neuropogon

trachycarpus

AUTHOR(S): Bodo, Bernard; Molho, Darius

CORPORATE SOURCE: Lab. Chim., Mus. Natl. Hist. Nat., Paris, 75005, Fr. SOURCE:

Phytochemistry (Elsevier) (1980), 19(6), 1117-20

CODEN: PYTCAS; ISSN: 0031-9422

DOCUMENT TYPE: Journal

LANGUAGE: French

The structures of the aliphatic acids, isomuronic (I; R = Ac) and neuropogolic acid (I; R = CHOHMe), isolated from N. trachycarpus, were determined by chemical and spectral means. CD allowed the configuration of isomuronic acid to be assigned as 2R.

70579-66-7 75716-00-6

RL: BIOL (Biological study)

(from Neuropogon trachycarpus, structure of)

70579-66-7 CAPLUS RN

3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-(14-oxopentadecyl)-, CN (2R)- (CA INDEX NAME)

Absolute stereochemistry.

75716-00-6 CAPLUS RN

CN 3-Furancarboxylic acid, 2,5-dihydro-2-(14-hydroxypentadecyl)-4-methyl-5oxo- (9CI) (CA INDEX NAME)

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

RN 75696-34-3 CAPLUS

CN 3-Furancarboxylic acid, 2-[14-[(2,4-dinitrophenyl))hydrazono]pentadecyl]-2,5-dihydro-4-methyl-5-oxo-, (R)- (9CI) (CA INDEX NAME)

Absolute stereochemistry. Double bond geometry unknown.

L3 ANSWER 25 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1980:617913 CAPLUS

93:217913 DOCUMENT NUMBER:

ORIGINAL REFERENCE NO.: 93:34751a,34754a TITLE: Lichen constituents. Part 123. Chemistry of some

yellow Acarospora species

AUTHOR(S): Huneck, S.

Inst. Biochem., DAW, Halle/Saale, DDR-401, Ger. Dem.

Rep.

SOURCE: Lichenologist (1980), 12(2), 239-42

CODEN: LCHNB8; ISSN: 0024-2829

DOCUMENT TYPE: Journal

LANGUAGE: English

Fifteen specimens of 4 Acarospora species of subgenus Xanthothallia were analyzed. All species contained (+)-rhizocarpic acid. A. gobiensis And A. schleicheri had only this compound, and A. oxytona this and (+)-lichesterinic acid. A. chlorophana Seems to exist in 2 chemical races,

one with a mixture of (-)-acaranoic and (-)-acarenoic acids and the other with (+)-roccellic acid. The stereochem. and biogenesis of these compds.

is briefly discussed.

ΤТ 70579-62-3

RL: BOC (Biological occurrence); BSU (Biological study, unclassified); BIOL (Biological study); OCCU (Occurrence)

(of Acarospora) 70579-62-3 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2R)- (CA INDEX NAME)

L3 ANSWER 26 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1980:579563 CAPLUS DOCUMENT NUMBER: 93:179563

ORIGINAL REFERENCE NO.: 93:28463a,28466a

TITLE: Anti-tumor activities of some lichen products and

their degradation products

AUTHOR(S): Hirayama, Teruhisa; Fujikawa, Fukujiro; Kasahara, Toshiko; Otsuka, Masako; Nishida, Noriko; Mizuno,

Denichi

CORPORATE SOURCE: Kvoto Coll. Pharm., Kvoto, Japan

SOURCE: Yakugaku Zasshi (1980), 100(7), 755-9

CODEN: YKKZAJ; ISSN: 0031-6903

DOCUMENT TYPE: Journal

LANGUAGE: Japanese

Anionic and cationic resins-adsorbed fractions of 44 lichens, hot water exts. of 9 lichens, and 20 lichen metabolites and their degradation products were assayed for their antitumor activity against ascitic or solid-type Ehrlich carcinoma. Among them, the adsorbed fraction of Ramalina almquistii , d-protolichesterinic acid [1448-96-0] and nephrosterinic acid [570-13-8] were effective against the solid-type Ehrlich carcinoma.

70579-62-3 75232-40-5 RL: BAC (Biological activity or effector, except adverse); BSU (Biological study, unclassified); THU (Therapeutic use); BIOL (Biological study); USES (Uses)

(antitumor activity of, from lichen)

70579-62-3 CAPLUS CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2R)- (CA INDEX NAME)

## Absolute stereochemistry.

RN 75232-40-5 CAPLUS

3-Furancarboxvlic acid, 2,5-dihvdro-4-methvl-5-oxo-2-undecvl- (9CI) (CA CN INDEX NAME)

L3 ANSWER 27 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1980:124916 CAPLUS DOCUMENT NUMBER: 92:124916

92:20329a,20332a ORIGINAL REFERENCE NO.:

Three new aliphatic acids from lichens of genus TITLE:

Parmelia (subgenus Xanthoparmelia) AUTHOR(S): Chester, Douglas O.; Elix, John A.

Dep. Chem., Aust. Natl. Univ., Canberra, 2600,

Australia

SOURCE: Australian Journal of Chemistry (1979), 32(11), 2565-9

CODEN: AJCHAS; ISSN: 0004-9425

DOCUMENT TYPE: Journal

LANGUAGE: English

The aliphatic acids, constipatic (I), protoconstipatic (II), and AB dehydroconstipatic (III), were identified as constituents of various Xanthoparmelia lichens from Australia.

72960-05-5 73036-28-9 RL: BIOL (Biological study) (from Xanthoparmelia)

72960-05-5 CAPLUS RN

3-Furancarboxvlic acid, 2,5-dihvdro-4-methvl-5-oxo-2-(14-oxopentadecvl)-, (S)- (9CI) (CA INDEX NAME)

Absolute stereochemistry.

RN 73036-28-9 CAPLUS

3-Furancarboxvlic acid, 2,5-dihvdro-2-(14-hvdroxvpentadecvl)-4-methvl-5-CN oxo- (9CI) (CA INDEX NAME)

L3 ANSWER 28 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1979:435683 CAPLUS

DOCUMENT NUMBER: 91:35683

ORIGINAL REFERENCE NO.: 91:5803a,5806a

TITLE: Neodihydromurol and murolic acid, two new

γ-lactonecarboxylic acids from Lecanora muralis
AUTHOR(S): Huneck, Siegfried; Schreiber, Klaus; Hoefle, Gerhard;

Snatzke, Guenther
CORPORATE SOURCE: Inst. Biochem., DAW, Halle/Saale, DDR-401, Ger. Dem.
Rep.

SOURCE: Journal of the Hattori Botanical Laboratory (1979),

45, 1-23

CODEN: JHBLAI; ISSN: 0073-0912

DOCUMENT TYPE: Journal

LANGUAGE: German

AB Two new aliphatic hydroxy \gamma-lactone carboxylic acids,

(+)-neodihydromurolic acid and (+)-murolic acid, were isolated from the lichens Lecanora muralis, L. melanophthalma, and L. rubina.

Spectroscopical and chemical data led to the following structures:

(+)-neodihydromurolic acid, (+)-2(S)-methy-3(S)-carboxy-4(R),18(R)-

dihydroxynonadecan-1→4-olide (I); and (+)-murolic acid, (+)-2-methylen-3(S)-carboxy-4(R),18(R)-dihydroxynonadecan-1→4-olide

(II). The absolute configurations of (+)-nephrosteranic acid,

(-)-alloprotolichesterinic acid, and (+)-nephrosterinic acid were established.

IT 70579-62-3P 70579-64-5P 70579-66-7P

70579-68-9P

RL: SPN (Synthetic preparation); PREP (Preparation)

(preparation of)

RN 70579-62-3 CAPLUS

3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2R)- (CA INDEX NAME)

Absolute stereochemistry.

RN 70579-64-5 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-2-[(14R)-14-hydroxypentadecyl]-4-methyl-5-oxo-, (2R)- (CA INDEX NAME)

Absolute stereochemistry.

RN 70579-66-7 CAPLUS

Absolute stereochemistry.

RN 70579-68-9 CAPLUS

CN 3-Furancarboxylic acid, 2-[14-(acetyloxy)pentadecyl]-2,5-dihydro-4-methyl-5-oxo-, [R-(R\*,R\*)]- (9CI) (CA INDEX NAME)

Absolute stereochemistry.

L3 ANSWER 29 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1977:546475 CAPLUS DOCUMENT NUMBER: 87:146475

ORIGINAL REFERENCE NO.: 87:23117a,23120a

TITLE: Effect of a group of cyclopentane naphthenic

derivatives on the permeability of blood capillaries

in animals

AUTHOR(S): Maizelis, M. Ya.; Kruglikov, R. I.; Omarov, I. A.

CORPORATE SOURCE: Azerb. Gos. Univ. im. Kirova, Baku, USSR

SOURCE: Uchenye Zapiski - Ministerstvo Vysshego i Srednego

Spetsial'nogo Obrazovaniya Azerbaidzhanskoi SSR, Seriya Biologicheskikh Nauk (1976), (1), 39-45

CODEN: UZMBDL; ISSN: 0132-7038

DOCUMENT TYPE: Journal

LANGUAGE: Russian
AB I.m. injections of a cyclopentane naphthenic acid (150 mg/kg), a

cyclopentame perhydrophenanthrenic naphthenic hydrocarbon (150 mg/kg), or lichesterinic acid [493-47-0] (5 mg/kg) for 10 days increased the vascular permeability of PO43- in the capillaries of rats from the

the vascular permeability of P043- in the capillaries of rats from the blood to tissue; however, sarcomycin [11031-48-4] had the opposite effect. In all cases vascular permeability was nearly normalized 10 days following completion of the various treatments.

IT 493-47-0

RL: PRP (Properties)

(capillary permeability increase by)

RN 493-47-0 CAPLUS

L3 ANSWER 30 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1972:511076 CAPLUS DOCUMENT NUMBER: 77:111076

ORIGINAL REFERENCE NO.: 77:18307a,18310a

TITLE: Separation and detection of lichesterinic acids by

thin-layer chromatography

AUTHOR(S): Kowalska, Maria

CORPORATE SOURCE: Wyzsza Szk. Roln., Poznan, Pol.

SOURCE: Roczniki Wyzszej Szkoly Rolniczej w Poznaniu (1971),

52, 15-22 CODEN: RWSPA2; ISSN: 0370-8020

CODEN: RWSPA2; ISSN: 0370-8020
DOCUMENT TYPE: Journal

DOCUMENT TYPE: Journa. LANGUAGE: Polish

AB A group of lichesterinic acids from Cetraria islandica and Usnea dasypoga

was studied by thin-layer chromatog. The compds. were separated on silica gel or polyamide by using either a system consisting of

CHCl3-MeOH-EtCOMeacetylacetone (20:10:5:1) or CHCl3-Me2CO-EtOH (8:2:2). The individual compds. were determined with 1% FeCl3 in MeOH.

IT 493-47-0D, 3-Furancarboxylic acid,

2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, derivs. RL: ANT (Analyte); ANST (Analytical study)

(detection of, in plant material, chromatog.)

RN 493-47-0 CAPLUS

L3 ANSWER 31 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1970:506362 CAPLUS DOCUMENT NUMBER: 73:106362

ORIGINAL REFERENCE NO.: 73:17307a,17310a

TITLE: Biosynthesis of (+)-protolichesterinic acid in

Cetraria islandica

AUTHOR(S): Bloomer, James L.; Eder, W. R.; Hoffman, William Freeman

CORPORATE SOURCE: Dep. of Chem., Temple Univ., Philadelphia, PA, USA SOURCE: Journal of the Chemical Society [Section] C: Organic

(1970), (13), 1848-50

CODEN: JSOOAX; ISSN: 0022-4952

DOCUMENT TYPE: Journal

LANGUAGE: English

Biosynthesis of (+)-protolichesterinic acid was studied by use of [1-14C]acetate and [1,4-14C2]succinic acid. The results support the hypothesis that aliphatic lichen acids have common precursors related to the citric acid and fatty acid cycles; however, the extremely low levels of incorporation suggest that the biosynthesis represents very minor metabolic pathways in C. islandica. The biosynthesis appears to be

inoperative in winter. 493-47-0P

RL: SPN (Synthetic preparation); PREP (Preparation) (preparation of)

493-47-0 CAPLUS

L3 ANSWER 32 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1969:77124 CAPLUS

DOCUMENT NUMBER: 70:77124

ORIGINAL REFERENCE NO.: 70:14369a,14372a
TITLE: Naturally occurring lactones and lactams. I.

Absolute configuration of ranunculin, lichesterinic acid, and some lactones related to lichesterinic acid

AUTHOR(S): Boll, Per M.

CORPORATE SOURCE: Univ. Copenhagen, Copenhagen, Den.

SOURCE: Acta Chemica Scandinavica (1947-1973) (1968), 22(10),

3245-50

CODEN: ACSAA4; ISSN: 0001-5393

DOCUMENT TYPE: Journal

DOCUMENT TYPE: Journal LANGUAGE: English

AB N.M.R. spectra have confirmed the provisional structure of ranunculin.

Circular dichroism data allowed the assignment of the configuration of its

aglucone to be 4S. As a result of the circular dichroism work, it was also possible to allocate configurations to the following lichen lactones: (S)-(-)-lichesterinic acid, (3R,4S)-(-)-protolichesterinic acid, 3r.4S)-(-)-protolichesterinic acid

(3S, 4S)-(-)-alloprotolichesterinic acid, and (2R, 3S, 4S)-nephromopsic acid.

IT 22800-25-5

RL: PRP (Properties)

(configuration of, absolute)

RN 22800-25-5 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2S)- (CA

INDEX NAME)

Absolute stereochemistry.

L3 ANSWER 33 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1967:497597 CAPLUS 67:97597

DOCUMENT NUMBER:

ORIGINAL REFERENCE NO.: 67:18339a,18342a

TITLE: Lichens. IV. Thin-layer chromatography of lichen

substances AUTHOR(S): Santesson, Johan

CORPORATE SOURCE: Univ. Uppsala, Uppsala, Swed.

SOURCE: Acta Chemica Scandinavica (1947-1973) (1967), 21(5), 1162-72

CODEN: ACSAA4; ISSN: 0001-5393

Journal

DOCUMENT TYPE: LANGUAGE: English

AB cf. CA 67: 51056p. The thin-layer chromatography on precoated plates of >80 lichen substances is described. 32 references.

ΙT 493-47-0

RL: ANT (Analyte); ANST (Analytical study) (thin-layer chromatog. of)

RN 493-47-0 CAPLUS

L3 ANSWER 34 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1966:475198 CAPLUS

DOCUMENT NUMBER: 65:75198 ORIGINAL REFERENCE NO.: 65:14079a-b

Lichens, II. Thin-layer chromatography of aliphatic TITLE:

lichen acids AUTHOR(S): Bendz, Gerd; Santesson, Johan; Tibell, Leif

Univ. Uppsala, Swed.

SOURCE: Acta Chemica Scandinavica (1966), 20(4), 1180-1 CODEN: ACHSE7; ISSN: 0904-213X

DOCUMENT TYPE: Journal

LANGUAGE: English

cf. CA 64, 13073b. Aliphatic lichen acids were separated by thin layer chromatog, on silica gel HF, by using 40 mg. bromcresol green in 100 mL. 0.01N NaOH as the detection spray. Rf values were tabulated.Rf + 100 in solvent system, A, B, C, D; Caperatic acid, 03, 02, 01, 11; Lichesterinic acid, 73, 32, 56, X; Nephromopsinic acid, 82, 32, 54, X; Nephrosteranic acid, 82, 31, 55, X; Nephrosterinic acid, 61, 22, 43, X, Norrangiformic acid, 04, 03, 03, 49; Acaranoic acid, 68, 26, 42, X; Acarenoic acid, 48, 17, 30, X; Protolichesterinic acid, 61, 23, 43, X; Rangiformic acid, 50, 10, 36, 66; Roccellic acid, 91, 24, 60, X; X indicates that the acid travels with the secondary front; the solvents were: (A) ether-butyric acid 20:1, (B) CHCl3-propionic acid 20:1, (C) iso-Pr ether-propionic acid 20: 1, (D) CHC13-HOAc 5:1.

493-47-0, Fumaric acid, (1-hydroxytetradecyl)methyl-,

γ-lactone

RN

(chromatog. of) 493-47-0 CAPLUS

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L3 ANSWER 35 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN
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ACCESSION NUMBER: 1958:113136 CAPLUS

DOCUMENT NUMBER: 52:113136

ORIGINAL REFERENCE NO.: 52:19935q-i,19936a-i,19937a-h

TITLE: The synthesis of dl-protolichesterinic acid
AUTHOR(S): Van Tamelen, Eugene E.; Bach, Shirley Rosenberg

CORPORATE SOURCE: Univ. of Wisconsin, Madison

SOURCE: Journal of the American Chemical Society (1958), 80,

3079-86

CODEN: JACSAT; ISSN: 0002-7863

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

OTHER SOURCE(S): CASREACT 52:113136

AB Me dl-dihydroprotolichesterinate (180 mg.), 0.024 g. Na, and 5.5 cc. MeOH refluxed 1 hr., poured into H2O, acidified with NaHSO4, extracted with Et2O, the extract worked up, the residue (0.129 g.) dissolved in 7 cc. MeOH, the solution treated with 1 cc. H2O containing 0.0304 g. NaOH, kept 5 days at room temperature, diluted with H2O, acidified with NaHSO4, and the precipitate

recrystd. from
glacial AcOH, washed with petr. ether, and recrystd. again from MeOH
yielded 0.056 g. neodihydroprotolichesterinic acid (I), platelets, m.
97-8° (all m.ps. are corrected) I with CRI2N2 gave the Me ester, m.
38-9° (uncor.). Me dl-isodihydroprotolichesterinate (0.31 g.) and
10.5 cc. absolute MeOH refluxed 5.5 hrs. with 0.00419 g. Na, treated with 1
cc. H2O, refluxed 6.5 hrs., cooled, diluted with H2O, acidified with NaHSO4,

cold petr.

ether left 0.070 g. I. C13H27COCH2CO2Me (II) (5 g.) and 2.9 g. powdered NaI added to 0.41 g. Na in 10 cc. absolute MeOH, the mixture treated with cooling during 10 min. with 3.0 g. BrCH2CO2EEt, kept 2 days at room temperature, filtered, the residue washed with H2O, the filtrate poured into H2O, acidified and extracted with Et2O, and the extract worked up yielded 2.53 g. dialkylation product, C25H44O7, m. 42-3°. II (10 g.), 100 cc. dry C6H6, and 10 g. pyrrolidine, b. 86.5-87° refluxed 9 hrs. with the azectropic removal of about 0.8 cc. H2O and evaporated gave 11.5 g. pyrrolidine enamine (III) of II, yellow liquid. III (11.5 g.), 100 cc. absolute M6OH, and 5.85 g. BrCH2CO2Et refluxed 29 hrs., and stirred overnight with 20 cc. H2O, the aqueous layer extracted with Et2O, and the combined organic

extracted with Et20, the extract worked up, and the residue extracted with

layer and extract evaporated gave 10 g. brown oily C13H27COCH(CO2Me)CH2CO2Et (IV); a 10-g, portion in 50 cc. absolute MeOH treated with 8 cc. 1.0M NaBH4 in MeOH, allowed to stand 3 days, treated again with 11 cc. NaBH4 solution, allowed to stand 3 hrs., poured into H2O, acidified with NaHSO4, and extracted with Et20, the extract washed, dried, and evaporated, the residual yellow oil dissolved with 7 g. KOH in 110 cc. 90% MeOH, allowed to stand 1 day at room temperature, cooled, filtered, the residue acidified with 5% HCl, digested 1 hr. at 70°, kept several hrs. at room temperature, filtered, dried (5.1 g.), and recrystd. from C6H6 yielded 4.8 g. 3-carboxy-4-oxoheptadecanoate (V), m. 80-3°. V (1 g.) treated with CH2N2 in Et2O and evaporated yielded 1.03 g.  $\beta$ -carbomethoxy- $\gamma$ -tridecyl- $\gamma$ -butyrolactone (VI), m. 68-70° (MeOH). (EtO)2CO (80 g.) and 8.6 g. butyrolactone refluxed at 125 mm., treated during 1 hr. with 2.39 g. Na in 56 cc. absolute EtOH while removing the EtOH simultaneously with the addition, the residual pale yellow, gelatinous mass poured into 60 cc. glacial AcOH and ice and extracted with 50 cc. Et20, and the extract worked up yielded 4.1 g. α-carbethoxy-γ-butyrolactone(VII), b0.5, 106-9°. VII in EtOH treated with excess liquid NH3 gave HO(CH2)2CH(CONH2)2, m. 152.5-53° (EtOH). VI (3 g.) and 7.55 g. (EtO)2CO treated dropwise during 1 hr. with stirring under reflux at 125 mm. with 0.212 g. Na in 5.6 cc. absolute EtOH while removing the EtOH continuously, the resulting slush poured into 6 cc. glacial AcOH and ice and extracted with Et20, and the extract worked up yielded 3.4 q. light red oil; a 0.79-q. portion chromatographed

on 12 g. silicic acid did not give the desired carbethoxylation product; a 2.37-g. portion in 20 cc. MeOH containing 1.27 g. KOH kept 5 days at room temperature, acidified with \$ HCl, filtered, and the residue washed with H2O, dried, and extracted with ligroine (b. 60-8°) left 1.4 g. material C18H32O4, m. 133-5°. C13H27CH:CHCO2H (VIII), m. 47-9° (aqueous EtOH), was prepared by the method of Myers (C.A. 46, 1438q) and separated in

yield from the by-product C14H29CH(OH)CO2H by extracting the crude mixture with petr. ether at room temperature, filtering, cooling to 0°, filtering again, evaporating, and recrystg. the residue from aqueous MeOH. VIII (5 g.)

cc. Et20 treated with CH2N2 in Et20 until the yellow color persisted for 5 min. and evaporated on the steam bath gave 5.3 g. Me ester (IX) of VIII. trans-VIII (1.0 g.) in a few cc. CC14 treated with about 8 cc. 58 CC14-Br in small portions during 0.5 hr. and evaporated, the residual yellow oily paste dissolved in 10 cc. Ac2O, the solution treated with 0.5 g. powdered KOAc, refluxed 3 hrs., treated with iced H2O, and filtered, the residual creamy paste refluxed 0.5 hr. with 15 cc. 6% alc. KOH, the mixture cooled, poured onto 50 g. ice containing a slight excess of dilute H2SO4, and extracted with Et2O,

the extract evaporated, and the residual pale yellow waxy solid triturated during several days at room temperature with a few cc. petr. ether gave 0.04 g.

compound

A, m. 88.5-9.5°; the filtrate from the isolation of compound A cooled
in ice gave 0.30 g. impure compound B, m. 56-61.5°; the crude compound
B treated with three 10-cc. portions ligroine at room temperature, the combined
exts. concentrated to 10 cc., cooled to 15°, and centrifuged, and the
precipitate washed with a little cold ligroine and recrystd. from ligroine at
10° yielded 10 mg. pure cis-2,3-epoxyhexadecanoic acid, flakes, m.
70.0-70.9°. (CF3CO)20 (21.2 cc.), 3.5 cc. 90% H2O2, and 25 cc.
CH2C12 added with cooling dropwise during 40 min, to 10.6 g. IX. 56.5 g.

CH2C12 added with cooling dropwise during 40 min. to 10.6 g. IX, 56.5 g. Na2HPO4, and 70 cc. dry CH2C12, refluxed 0.5 hr., and stirred with 100 cc. H2C0, the aqueous layer washed with 70 cc. CH2C12, and the combined organic layer

and extract washed, dried, and worked up yielded Me tridecylglycidate (X) in 3 fractions: (1) b0.4 140-6°, 3.73g.; (2) b0.4 148-50°, 2.62 g.; (3) b0.4 150-2°, 3.73 g. X (0.2902 g.), 10 cc. dioxane, and 0.5 cc. 10% aqueous NaOH refluxed 1.5 hrs. under N, cooled, poured into iced H2O containing 5 cc. 5% HCl, and extracted with Et2O, the extract worked up, and

residual oil diluted with 8 cc. petr. ether, cooled, and filtered yielded 0.122 g. trans-tridecylglycidic acid, platelets, m. 86-7°. Na (0.485 g.) in 8 cc. absolute MeOH treated with 2.79 g. CH2(CO2Me)2, the mixture treated during 10 min. with stirring with 6.00 g. X in 10 cc. absolute MeOH, refluxed 4 hrs., cooled, poured into 150 cc. ice and H2O, acidified with 5% HCl, extracted with CHCl3, and the extract worked up gave 7.85 g. crude, pale

yellow, oily product which chromatographed on silicic acid gave pure  $\alpha,\beta-$  dicarbomethoxy-y-tridecyl-y-butyrolactone (XI), white wax. XI (2.1 g.) in 40 cc. MeOH treated with 5 cc. H2O containing 1.84 g. KOH, refluxed 3 hrs., kept overnight at room temperature, decanted, the oily residue dissolved in 50 cc. H2O, the solution acidified with 5% HCl to Congo red and filtered, and the residue dried (1.182 g.) and recrystd. from 20 cc. hot MeOH yielded 0.721 g. mono-K salt (XII) of  $\alpha,\beta$ -dicarboxy-y-tridecylbutyrolactone (XIII), powder, m.  $124^{\circ}$  (decomposition); the mother liquor poured into 100 cc. H2O, acidified with 5% HCl, extracted with Et2O, and the extract worked up gave

g, white material. XII (0.0394 g.) refluxed 0.5 hr. with 0.5 cc. 5% H2SS4, cooled, extracted with Et2O, and the extract worked up gave 0.0265 g. mixed disstereoisomers of  $V, m.~87.5-94.5^\circ$ . XII (0.050 q.) in 5

cc. MeOH acidified with 5% HCl, diluted with H2O, extracted with Et2O, and the extract dried and evaporated under N at room temperature gave 0.036 g. XIII. (0.372 g.) treated with 0.207 g. Et2NH and 0.126 g. 30% agueous CH2O, diluted

with 2 cc. MeOH, heated 1 min. on the steam bath, kept 1 day at room temperature, treated again with 0.126 g. 30% aqueous CH2O, allowed to stand 1

day, diluted with a few cc. MeOH, evaporated, the residue evaporated twice with CHC13.

the resulting solid kept overnight in 5 cc. CHC13 and filtered, and the residue (0.114 g.) dissolved in glacial AcOH, treated with a few drops H2O, cooled to 15°, and filtered gave 0.061 g.

dl-protolichesterinic acid (XIV), m. 92.5-4.5 $^{\circ}$  the filtrate from the crude XIV K salt evaporated, the residual semisolid dissolved in 2 cc. dry C6H6, the solution kept 3 days at room temperature with 5 cc. MeI, filtered, evaporated

at about 40° under N, the residual crude oil (0.338 g.) dissolved in 4 cc. MeOH, the solution treated with 5.5 cc. 5% aqueous NaHCO3, allowed to stand 3 days, diluted with H2O, extracted with Et2O, the aqueous solution acidified

with 5% HCl and extracted with Et2O, and the extract worked up yielded 0.0513  $\sigma$ .

(crude) XIV, m. 87.5-97.5°. Crude XIV (74 mg.) chromatographed on 5 g. silicic acid gave 29% purified d1-lichesterinic acid (XV), m. 114-15°, 42% XIV, m. 100.5-101.5°, and 11.8% less pure XIV, m. 98.5-100°. XIV (30 mg.) and 5 cc. Ac20 heated 1 hr. on the steam bath, cooled, diluted with H2O, and filtered yielded 21 mg. XV, m. 113-15° (AcOH). XIV (20 mg.) in 10 cc. glacial AcOH hydrogenated over 50 mg. 10% Pd-C, filtered, diluted with H2O, the precipitate recrystd.

from

XII

AcOH, and the product extracted with boiling ligroine and recrystd. from AcOH yielded 9 mg. dihydro derivative of XV, m. 114-16°. XII (0.3835 g.), 3 cc. MeOH, 0.079 g. Me2NH.HCl, 0.0873 g. Me2NH, and 0.097 g. 30% aqueous CH2O kept 2 days at room temperature, filtered, treated with a few cc. MeOH, evaporated

in vacuo on the steam bath, this procedure repeated twice with the addition and removal of CHCl3, the residual waxy solid treated with 3 cc. dry C6H6 and 5 cc. MeI, the mixture kept 3 days at room temperature, filtered, and the residue (0.653 g.) recrystd. from glacial AcOH yielded 0.340 g. methiodide (XVI), platelets, m. 165° (decomposition); the filtrate evaporated under N, the residual yellow oil (0.126 g.) dissolved in 2 cc. MeOH, the solution treated 3 days at room temperature with 2.1 cc. 5% aqueous NaHCO3 and extracted with

Et20, the aqueous phase acidified with 5% HCl and extracted with Et20, the extract

dried and evaporated, and the residue (0.038 g.) extracted with ligroine and recrystd. from aqueous AcOH gave 0.010 g. V, m. 98-100°. MeOH (5 cc.) and 2.8 cc. 5% aqueous NaHCO3 added to 0.211 g. XVI, kept 3 days at room temperature, diluted with H2O, washed with CHCl3, acidified, extracted with CHCl3, and

the extract worked up yielded 0.029 g. XIII, m. 92-5° (AcOH).

T 493-47-0P, Lichesterinic acid RL: PREP (Preparation)

(preparation of) 493-47-0 CAPLUS

$$\begin{array}{c} \text{O} & \text{(CH2)}_{12}\text{-Me} \\ \text{Me} & \text{CO}_2\text{H} \end{array}$$

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L3 ANSWER 36 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER:
                        1957:34629 CAPLUS
DOCUMENT NUMBER:
                        51:34629
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ORIGINAL REFERENCE NO.: 51:6517c-i,6518a-d

TITLE: Preparation and properties of the isomeric forms of

α-amino- and α.ε-diaminopimelic

acid

AUTHOR(S): Wade, Roy; Birnbaum, Sanford M.; Winitz, Milton;

Koegel, Robert J.; Greenstein, Jesse P.

CORPORATE SOURCE: Natl. Insts. of Health, Bethesda, MD

SOURCE: Journal of the American Chemical Society (1957), 79,

648-52 CODEN: JACSAT; ISSN: 0002-7863

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

OTHER SOURCE(S): CASREACT 51:34629

CH2(CH2CO2Et)2 cyclized by the method of Dobson, et al., (C.A. 4, 1028) yielded 76% α-carbethoxycyclohexanone (I), b0.4 70-2°. I

coupled with PhN2C1 by the method of Jackson and Manske (C.A. 25, 514) gave 60% Et H α-oxopimelate phenylhydrazone, m. 141-2°

(decomposition), which saponified with 1.1N NaOH in 50% aqueous dioxane gave HO2C(CH2)4C(:NNHPh)CO2H (II), prisms, m. 141-3° (decomposition) (from EtOAc-petr. ether). II (10 g.) refluxed 6 hrs. with 15 g. Zn dust and 150 cc. 75% AcOH, filtered, and evaporated, the residue dissolved in 50 cc. H2O, treated 3 hrs. with H2S; filtered hot, and evaporated to dryness, and the

crystalline residue shaken with a little EtOH and filtered gave HO2C(CH2)4CH(NH2)CO2H (III), plates, m. 216° (decomposition) (from aqueous

EtOH). III (3.5 g.) in 25 cc. 2N NaOH treated at 5° with 2.2 cc. Ac20 and 20 cc. 2N NaOH in alternate portions with shaking and cooling,

the mixture kept 1 hr. at room temperature, acidified to about pH 1.7 with 4N HC1 and evaporated at  $40^{\circ}$  in vacuo, the residue diluted with 20 cc. H2O, the

evaporation repeated, the crsvt. residue extracted with hot Me2CO, and the extract filtered, concentrated, diluted with Et20 to incipient turbidity, scratched,

and filtered yielded 2.5 q. N-Ac derivative (IV) of III, m. 111-12° (from Me2CO-Et2O). IV (2.5 g.) in 100 cc. H2O adjusted to pH 7.0-7.5 with 2N LiOH, treated with 1 g. renal acylase I, diluted to 130 cc., incubated about 4 hrs. at 39°, concentrated to 50 cc. in vacuo, dialyzed 4 times against

750 cc. H2O, the combined dialyzates (3 l.) concentrated to 15 cc. in vacuo, adjusted to pH 3.4 with 6N HCl, concentrated to beginning crystallization, diluted with 50

cc. absolute EtOH, and kept 24 hrs. at room temperature gave 800 mg. L-III,  $[\alpha]D26$  21.5° (c 1, 5N HCl); the filtrate acidified to pH 1.7, evaporated to dryness in vacuo, and extracted with boiling Me2CO, the extract concentrated

in an air stream, the residual oil refluxed 2 hrs. with 125 cc. 2N HCl and evaporated to dryness in vacuo, the residue dissolved in a little H2O, the pH adjusted to 3.4 with 2N LiOH, and the solution concentrated to beginning

crystallization and diluted with absolute EtOH yielded 500 mg. D-III, [a]D26 -21.0° (c 1, 5N HCl). D- and L-III gave the following Rf values (developer, and paper given): 0.44, PhOHNH4OH, Whatman Number 4; 0.43, 4:1:5 BuOH-AcOH-H2O, Whatman Number 4: 0.73, 10:77:20 pyridine-MeOH-H2O, Whatman Number 1. A

mixture

of the 3 isomers of CH2[CH2CH(NH2) CO2H]2 (V) was prepared in essentially the same manner in 66% yield; it showed 2 ninhydrin-sensitive spots with Rf values 0.46 and 0.57 corresponding to meso-V and D- and L-V. V (9.5 g.) in 125 cc. 2N NaOH treated with 19.5 cc. PhCH2OCOCl in portions with cooling and stirring during about 0.5 hr., the mixture shaken 2 hrs. at room temperature and washed with EtOAc, the aqueous layer acidified to pH 1.7 with HCl, the precipitated oil extracted into EtOAc, the extract dried, concentrated to  $50\,^{\rm o}$ 

in vacuo, kept at  $4^\circ$  overnight, and filtered, and the filter residue recrystd. from EtOAc gave 6.0 g. di(carbobenzyloxy) derivative (VI) of DL-V, m.  $164-5^\circ$  with shrinking at  $155^\circ$ . The combined EtOAc mother liquors from VI evaporated, and the qummy residue crystallized from hot

CHCl3 gave 6.2 g. meso-isomer (VII) of VI, m 123-5°. VII (30 g.) in 300 cc. AcOH and 100 cc. H2O hydrogenated over Pd black, filtered,

concentrated in vacuo, diluted with 50 cc., evaporated again, and recrystd. twice from

35% agueous EtOH yielded 7.5 g. meso-V, Rf 0.45. VI (45.8 g.) and 27.8 cc. Et3M in 600 cc. dioxane treated slowly with cooling with 24.4 cc. iso-BuCOC1 below 12°, kept 1 hr. at 10°, treated dropwise with 26 cc. NH4OH(d. 0.90), allowed to stand 16 hrs., and filtered by suction yielded 18.0 g. diamide (VIII) of VI, mass of needles, m. 223-4° (from aqueous HCONMe2). VIII (21.5 g.) hydrogenolyzed in 400

cc. AcOH over Pd black, filtered, evaporated, diluted with 25 cc. H2O, and again

evaporated, the residual oil dissolved in 300 cc. H2O containing 1.15 q. Mn(OAc)2.4H2O, the pH adjusted to 6.5 with 2N LiOH, the mixture treated with 1.8 g. lyophilized amidase powder, the pH adjusted to 8.0 with 2N LiOH, diluted to 470 cc., kept 5 hrs. at 38°, concentrated to about 50 cc., dialyzed 4 times against H2O (about 900 cc. each time) at 5°, the combined dialyzates concentrated to about 50 cc. in vacuo, passed through Amberlite XE-64 (Li+ form), and collected in 20-cc. fractions, the combined fractions 19-31 evaporated to dryness, the residue dissolved in the min. amount of hot H2O, the solution treated with C, filtered, adjusted to pH 6.5 with 2N LiOH, and diluted with 4 vols. absolute EtOH, and the white amorphous precipitate reppted. twice in the same manner yielded 3.5 g. L-V, Rf 0.57, [α]D26 45.0° (c 1, N HCl). The fractions from number 176 on combined and evaporated in vacuo, the residual sirup refluxed 6 hrs. with 1 1. 3N HCl, evaporated, dissolved in 1.5N HCl, and passed through Dowex 50, and the effluent adjusted to 2.5N HCl and evaporated gave 2.9 g. D-V,  $[\alpha]D26-45.5^{\circ}$  (c 1, N HCl). The infrared absorption spectra

of L-III, meso-V, and DL-V are recorded. 493-47-0P, Lichesterinic acid

RL: PREP (Preparation) (preparation of)

RN 493-47-0 CAPLUS

ΙT

L3 ANSWER 37 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1957:34628 CAPLUS

DOCUMENT NUMBER: 51:34628
ORIGINAL REFERENCE NO.: 51:6517b-c

TITLE: Synthesis of (±)-protolichesterinic acid

AUTHOR(S): Van Tamelen, E. E.; Bach, S. R. CORPORATE SOURCE: Univ. of Wisconsin, Madison

CORPORATE SOURCE: Univ. of Wisconsin, Madison
SOURCE: Chemistry & Industry (London, United Kingdom) (1956)

1308

CODEN: CHINAG: ISSN: 0009-3068

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

AB cf. C.A. 50, 6322a). A stereoselective synthesis of

(±)-protolichesterinic acid (I) was carried out. Me 2-hexadecenoate with CF3CO3H ylelded Me 2,3-epoxyhexadecanoate, b0.4 1H8-52° R.148-52° R.109 opening with di-Me malonate anion yielded, after spontaneous cyclization

of the intermediate γ-hydroxy ester,

 $\alpha$ , $\beta$ -dicarbomethoxy- $\gamma$ -n-tridecyl- $\gamma$ -butyrolactone. This on hydrolysis with hot MeOH-KOH was converted to the mono-K salt of the diacid, m. 124°, which with HCHO and Et2NH yielded I, m.

100.5-1.5°. Identification was confirmed by 3 separate tests. IT 493-47-0P, Lichesterinic acid

RL: PREP (Preparation)

(preparation of)

RN 493-47-0 CAPLUS

L3 ANSWER 38 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1956:36797 CAPLUS

DOCUMENT NUMBER: 50:36797 ORIGINAL REFERENCE NO.: 50:7242c-d

TITLE: Chemical components of Parmelia species of India

AUTHOR(S): Rangaswami, S.; Rao, V. Subba

CORPORATE SOURCE: Andhra Univ., Waltair

Indian Journal of Pharmacy (1955), 17, 50-3

CODEN: IJPAAO; ISSN: 0019-5472

DOCUMENT TYPE: Journal LANGUAGE:

Unavailable

Samples of P. nilgherrensis (I), P. perlata (II), and P. cirrhata (III) were examined All contained atranorin. Collatolic acid was found in I; II contained lecanoric acid; III contained d-protolichesterinic acid and

salazinic acid. IT

493-47-0, Fumaric acid, (1-hydroxytetradecyl)methyl-, γ-lactone

(in Parmelia)

RN 493-47-0 CAPLUS

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SOURCE:
                        Journal of the American Chemical Society (1955), 77,
                        4625-9
                        CODEN: JACSAT: ISSN: 0002-7863
DOCUMENT TYPE:
                        Journal
LANGUAGE:
                        Unavailable
GT
    For diagram(s), see printed CA Issue.
AB
     The Me ester (I) of d1-lichesterinic acid O.CO.CMe:C(CO2H).CH(CH2)12Me
     (II) has been synthesized by the SO2C12 dehydrogenation of Me ester (III)
     of dl-dihydroprotolichesterinic acid (IV), which was prepared by the NaBH4
     reduction of C13H27COCH(CO2Me)CHMeCO2Me (V). Various transformations
     encountered in the catalytic reduction of II and protolichesterinic acid
     (VI) are presented, and the possible biogenetic origins of these
     substances are discussed. C13H27COCH2CO2Me (VII), m. 38-9°, was
     prepared in 40% yield by the method of Stallberg-Stenhagen (C.A. 41, 4105d),
     filtering the crude product by suction with a rubber dam and recrystg. at
     0° from petr. ether. VII (5.0 g.), 2.9 g. NaI, and 3.18 g.
     MeCHBrCO2Et added to 0.41 g. Na in 10 cc. absolute MeOH, the mixture heated a
     few min. on the steam bath, held 4-7 days at room temperature, poured into H2O,
     acidified with NaHSO4, and filtered, and the waxy filter residue recrystd.
     from 30 cc. ligroine (b. 60-8°) gave 4.35 g. C13H27
     COCH(CO2Me)CHMeCO2Me (VIII), colorless prisms, m. 49-50°. VIII (5
     g.) in 50 cc. absolute MeOH held 3 days at room temperature with 3.9 cc. 1.0M
NaBH4
     in MeOH, the mixture treated with an addnl. 5.5 cc. NaBH4 solution, allowed to
     stand 3 hrs., and poured into H2O, the mixture acidified with NaHSO4, the
    precipitated oil extracted into Et20, the extract dried and evaporated, the
oilv residue
     refluxed 19 hrs. with 3.5 g. KOH in 55 cc. 90% MeOH, the precipitate filtered,
     dissolved in H2O, and acidified with 5% HCl, the crude precipitate extracted
with
     petr. ether, and the insol. residue recrystd. from glacial AcOH yielded
     1.70 q. IV, m. 114-15°; the filtrate of the hydrolysis mixture poured
     into a large excess H2O and acidified with NaHSO4, the crystalline precipitate
dried
     and extracted with boiling ligroine (b. 60-8°) to remove some II, m.
     84.5-5.0°, and the residue recrystd. from glacial AcOH yielded 9%
     dl-isodihydroprotolichesterinic acid (IX), m. 135-6°. IV treated
     with CH2N2 gave III, m. 62.0-2.5° (from MeOH). Similarly was
     prepared the Me ester of IX, m. 67.0-7.15°. d-VI hydrogenated in
    glacial AcOH at room temperature over 10% PdC, the mixture diluted with H2O,
and the
     precipitate recrystd. from glacial AcOH yielded 60% d-IV, m. 103.5-4.5°;
     Me ester, m. 54.5-5.5°. VI (1.8 g.) hydrogenated in the same
     manner gave dl-IV, m. 109-16°. C13H27CH:CHCO2H (8.8 g.) in 500 cc.
     H2O containing 18.5 g. KOH cooled to 0° with stirring, the resulting
     suspension warmed to room temperature, treated with stirring during 4 hrs. with
     2.50 g. Cl gas, and acidified with an equivalent amount H2SO4, the white solid
     precipitate dissolved in Et20, the solution dried and concentrated, the
residual pale
     yellow oil dissolved in 90 cc. ligroine, the solution cooled several days at
     0-5°, and the crystalline deposit (2.3 g.) recrystd. from ligroine gave
     1.7 g. chlorohydroxydecanoic acid, m. 75.7-6.2°; Et ester, m.
     50.8-1.5°. III (200 mg.), 160 mg. SO2C12, and 10 mg. Bz2O2 in 0.5
     cc. CC14 refluxed 18 hrs., the solvent removed in vacuo, the residue
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L3 ANSWER 39 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

50:31889

1956:31889 CAPLUS

Shirley Rosenberg

Univ. of Wisconsin, Madison

Synthesis of dl-lichesterinic acid methyl ester Van Tameslen, Eugene E.; Osborne, Clyde E., Jr.; Bach,

ACCESSION NUMBER:

ORIGINAL REFERENCE NO.: 50:6322a-i

DOCUMENT NUMBER:

CORPORATE SOURCE:

TITLE:

AUTHOR(S):

treated with H2O and 20 cc. Rt2O, the Rt2O layer dried and evaporated, the residue dissolved in 1 cc. EtOH, the solution filtered, and chilled, and the solid deposit dried and recrystd. from MeOH yielded 7-17% I, m. 49-50°. II (5 mg.) from equal parts of the optical antipodes treated with CH2N2 in Rt2O yielded I, m. 51-2°. IV heated with Br in polyphosphoric acid at 120-40° and the resulting product treated with clidine gave an unseparable mixture of products. IV treated with N-bromosuccinimide and Bz2O2 gave crude material containing about 7% II. dl-I (9.6 mg.) in 2 cc. MeOH treated with I cc. 2.66 + 10-2M aqueous NaOH, the solution held 5 days at room temperature, acidified with NaHSO4, and tered.

filtered,
the filter residue dissolved in ligroine, the solution filtered and evaporated,
and the residue recrystd. gave d1-11, m. 83-4°. d-II (540 mg.) in

200 cc. glacial AcOH hydrogenated over 200 mg. PtO2, the mixture filtered, the filtrate diluted with H2O, and the precipitate extracted with boiling ligroine and

recrystd. 3 times from glacial AcOH yielded 250 mg. C13H27CH(CO2H)CHMeCO2H (X), m. 135.5-6.5°. X (82 mg.) heated 1 hr. at 100° in a sealed tube with 0.4 cc. AcCl, the excess AcCl evaporated, and the residue recrystd. from ligroine, at -78° gave 57% anhydride of X, m.

34°. II 493-47-0P, Fumaric acid, (1-hydroxytetradecyl)methyl-, dl-,  $\gamma$ -lactone

RL: PREP (Preparation) (preparation of)

RN 493-47-0 CAPLUS

L3 ANSWER 40 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1954:78414 CAPLUS DOCUMENT NUMBER: 48:78414

ORIGINAL REFERENCE NO.: 48:13836b-d

TITLE: Chemical investigation of the lichens: Parmelia

kamtschadalis and Parmelia arnoldii

AUTHOR(S): Shah, Latika G.

CORPORATE SOURCE: Inst. Sci., Bombay

SOURCE:

Journal of the Indian Chemical Society (1954), 31,

253-6

CODEN: JICSAH; ISSN: 0019-4522

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

AB Solvent extraction of 2 varieties of lichen led to the recovery and identification of several crystalline substances. Air-dried Parmella

kamtschadalis (400 g.) was extracted with cold petr. ether, the extract was concentrated, and the material which separated was recrystd. from CHC13-EtOH to give

0.05 g. of atranorin (I), m. 195-7°. The material left after the

petr. ether extraction was repeatedly extracted with Et20. The concentrated extract gave  $\boldsymbol{2}$ 

g. I. The Bt20 filtrate was extracted with NaHCO3 solution Acidification and extraction of the aqueous solution with Bt20 and evaporation of the dried solution gave 1.0 q.

of protolichesteric acid (II), on crystallization from alc. m. 104-5°,

 $[\alpha]D = +9^{\circ}$  (7-9%, alc.). Lichesteric acid (III), m.

120-2°, crystallized from the diluted filtrate from the crystallization of II. The residue from the Et20 extraction of the lichen was extracted with alc. The extract was concentrated and yielded crystalline salazinic acid (IV). The alc. filtrate

was evaporated to dryness to give a sirupy mass containing a reducing sugar. Attempts to prepare an osazone were unsuccessful. Refluxing in Ac20 with pyridine gave a tetraacetate, m.  $68^{\circ}$ . Further extraction of the lichen with EtOAc gave an addnl. 1.0 g. of IV, while extraction with Me2CO gave 5.2 g. addnl. IV. Air-dried P. arnoldii (300 g.) extracted as described for P. kamtschadalis gave I and lecanoric acid,  $178-81^{\circ}$ , from the EtOA

extract The EtOAc extract gave IV. 493-47-0, Fumaric acid, (1-hydroxytetradecyl)methyl-, \gamma-lactone

(in Parmelia kampschadalis)

RN 493-47-0 CAPLUS

L3 ANSWER 41 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1954:15247 CAPLUS

DOCUMENT NUMBER: 48:15247
ORIGINAL REFERENCE NO.: 48:2822g-h

TITLE: The antibiotic action of lichen substances

AUTHOR(S): Klosa, Josef

CORPORATE SOURCE: Altheiderstr. 11, Berlin

OURCE: Hoppe-Seyler's Zeitschrift fuer Physiologische Chemie

(1951), 287, 197-204

CODEN: HSZPAZ; ISSN: 0018-4888

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

AB All of the 82 lichen substances tested had strong antibiotic action against Micrococcus pyogenes var. aureus, Streptococcus pyogenes, pneumococci, and diphtheria bacteria. The strongest antibiotic action was found in the Parmeliaceae, Cladoniaceae, and Usneaceae. Purified lichen acids also showed antibiotic properties. The in vitro antituberculous action of the lichen substances was reduced by the addition of serum.

IT 493-47-0, Fumaric acid, (1-hydroxytetradecyl)methyl-,

γ-lactone

(antibiotic action of)

RN 493-47-0 CAPLUS

L3 ANSWER 42 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1952:68533 CAPLUS DOCUMENT NUMBER: 46:68533

ORIGINAL REFERENCE NO.: 46:11463i,11464a

TITLE: d-Lichosteric acid-effect in vivo on pigmented mice

with inoculation tuberculosis

AUTHOR(S): Vartia, K. O.; Tervila, Leo

CORPORATE SOURCE: Univ. Helsinki, Finland
SOURCE: Annales Medicinae Experimentalis et Biologiae Fenniae,

Supplementum (1952), 30, 76-8

CODEN: AMBSA9; ISSN: 0066-2178
DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

AB Administration of d-lichesteric acid to mice infected with tuberculosis did not affect the course of the disease, while distinct retardation was observed if the latter was administered with streptomycin.

IT 493-47-0, Lichesteric acid

(effect on pigmented mice with inoculation tuberculosis)

RN 493-47-0 CAPLUS

L3 ANSWER 43 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1952:52941 CAPLUS DOCUMENT NUMBER: 46:52941

ORIGINAL REFERENCE NO.: 46:8811e-i,8812a

TITLE: Antibiotic effects of lichen and lichen substances

AUTHOR(S): Vartia, K. O. CORPORATE SOURCE: Helsinki Univ., Finland

Annales Medicinae Experimentalis et Biologiae Fenniae,

Supplementum (1950), 28(Suppl. 7), 5-82

CODEN: AMBSA9; ISSN: 0066-2178

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

In preliminary tests made with pieces of lichen, 75 out of 149 forms (50%) were found distinctly active towards a min. of 2 bacteria studied. Of these, the active substance of 50, or 2/3, was known. Gram-pos. bacteria only, as a rule, were susceptible; the distinct inhibitory effect on gram-neg, rods observed in some cases was obviously due to the decomposition products of lichen substances. Of the total of 20 crystalline lichen substances or related compds. 15, of different inhibitory profiles, proved to be more or less active against the rapidly growing gram-pos. bacteria and the tubercle bacillus(TB). The substances tested represented 8 types of lichen substances: the aliphatic lactones (d-protolichesteric and d-lichesteric acids) inhibited fairly strongly the growth of rapidly growing bacteria, in particular those of the aliphatic fatty acid type (lichesterylic and caperatic acids) revealing a comparatively better inhibitory effect on the growth of the TB, as did the pulvic acid derivs. (pinastric acid and the anilide of pulvic acid). The cumarone derivative (usnic acid) was of the same effective range as the most active lichen substances of other types. The activity of the depsides of orcinol type (evernic, divaricatic, gyrophoric, and umbilicaric acids) and that of the depsidones of orcinol type (physodic acid) seemed to increase with the growth in length of the side chains, except as regards the tubercule bacillus. The chlorine-containing diploicin was comparatively best in effecting gram-pos. dust bacteria. Two usnic acid derivs. only (usnolic and decarbousnic acids) and the depsidones of  $\beta$ -orcinol type (fumarprotocetraric and salazinic acids and the hexaacetate of salazinic acid) were found completely inactive. The depside of  $\beta$ -orcinol type (atranorin) also was very weakly active only against the rapidly growing bacteria, inhibiting the growth of the tubercle bacillus comparatively better. The decomposition product of atranorin (atranol) had a distinct inhibitory effect on the growth of gram-neg, bacteria. With some individual lichen substances of different types distinct activity on various fungal strains was observed. The nature of the different types of lichen substances seems to depend, apart from the basic structural formula of the substance, to a surprisingly great degree on seemingly insignificant changes in their mols.

493-47-0P, Lichesteric acid RL: PREP (Preparation)

(preparation of) RN 493-47-0 CAPLUS

L3 ANSWER 44 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1951:39033 CAPLUS

DOCUMENT NUMBER: 45:39033

ORIGINAL REFERENCE NO.: 45:6691h-i,6692a-b

TITLE: Antibacterial effects of lichen substances. I.

Comparative studies of antibacterial effects of various types of lichen substances

Shibata, Shoji; Miura, Yoshiaki; Sugimura, Hisako;

Tovoizumi, Yuri

CORPORATE SOURCE:

Univ. Tokvo SOURCE: Yakugaku Zasshi (1948), 68, 300-3 CODEN: YKKZAJ; ISSN: 0031-6903

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

cf. preceding abstract The relation between the chemical structure of usnic acid and its antibacterial effects described in previous papers was discussed. Comparatively powerful antibacterial activities against gram-pos. bacteria were found in lichesterinic acid and its derivs. and in depsides from orcinols having large alkyl radicals. No antibacterial activities were found in fatty acids of the caperatic acid type, depsides of the \$-orcinol series, depsidones, and endocrocin related to anthraquinone. None showed any activity against gram-neg, bacteria. highest dilns, inhibiting growth of M. tuberculosis (avian type) and Staph. aureus, resp., were: protolichesterinic acid -, 1:80,000; 1-lichesterinic acid 1:40,000, 1:160,000; 1-dihydroprotolichesterinic acid 1:80,000, 1:80,000; caperatic acid -, 1:5,000; rangiformic acid -, < 1:5,000; zeorin -, < 1:5,000; lecanoric acid -, < 1:5,000; divaricatic acid 1:10,000, 1:80,000; sphaerophorin -, 1:80,000; anziaic acid -, 1:80,000; perlatolinic acid 1:40,000, 1:80,000; olivetoric acid 1:10,000, 1:20,000; sekikaic acid 1:10,000, 1:80,000; ramalinolic acid -, 1:20,000; boninic acid -, 1:10,000; atranorin -, < 1:5,000; thamnolic acid -, < 1:5,000; lobaric acid -, 1:20,000; salazinic acid -, 1:5,000; psoromic acid -, 1:5,000; fumarprotocetraric acid -, < 1:5,000; pannarin -, < 1:5,000; endocrocin -, <1:5,000.

493-47-0, Lichesteric acid

(and derivs., antibacterial effects of)

RN 493-47-0 CAPLUS

CN

L3 ANSWER 45 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1949:6300 CAPLUS DOCUMENT NUMBER: 43:6300

ORIGINAL REFERENCE NO.: 43:1322b-f

TITLE: Lactone aliphatic acids as antibacterial agents AUTHOR(S): Cavallito, Chester J.; Fruehauf, Dorothy M.; Bailey,

John H. Journal of the American Chemical Society (1948), 70,

3724-6

CODEN: JACSAT: ISSN: 0002-7863

DOCUMENT TYPE: Journal

LANGUAGE:

Unavailable GI For diagram(s), see printed CA Issue.

A study has been made of the relationship between lactone structure and antibiotic activity. The Na salt of  $\alpha$ -carbethoxybutyrolactone (18 g.) in 250 cc. absolute EtOH and 0.1 mol. of the alkyl bromide were refluxed 4 hrs., the reaction mixture poured into 500 cc. H2O, extracted with three

150-cc.

portions of CHC13, and the residue saponified with 8.4 g. KOH in 150 cc. EtOH; the yields of the substituted  $\alpha$ -carboxybutyrolactones, H2C.CH2.CR(CO2H).CO.O, were from 20 to 45% (R is given): C10H21 m. 75-7° (m.ps. corrected), n (in 0.1 M K phosphate buffer at pH 7; acid concentration 3 + 10-5 millimol./cc.) 70.3; C12H25 m. 78-9°, ε 68.1; C13H27 m. 69-70°, η 43.3; C14H29 m. 82-3°,  $\eta$  35.0 ( $\gamma$ -Me derivative m. 64-7°,  $\eta$  33.2); C16H33 m.  $80-2^{\circ}$ ,  $\eta$  41.4 ( $\gamma$ -Me derivative m.  $60-3^{\circ}$ ,  $\eta$  37.6). 1-Protolichesterinic acid (I) (1.5 g.) and 1.5 g. 1-cysteine-HCl in dilute NaHCO3 (pH 7), kept 20 hrs. at 25° and the solution strongly acidified with HCl, give 1 g. of the 1-cysteine derivative

of I, m. 185-8° (decomposition); the addition appears to be through the SH group. Data are given for the min. bacteriostatic concentration for Streptococcus hemolyticus C203, Staphylococcus aureus 209, Clostridium welchii, Bacillus typhi, and B. tuberculosis ranae and H37Rv for the above lactones, I, II, 1-lichesterinic acid, 1-dihydroprotolichesterinic acid, and chaulmoogric acid. The antibacterial activity of I is related to its effect on  $\eta$  and not to any significant extent on the unsatd. system. II is much less inhibitory to bacteria than is I. Of the lactones, the C14 chain was optimum in contributing to the antibacterial activity and the  $\gamma-\text{Me}$  derivative has about the same activity. The lactone aliphatic acids are more compatible with complex media than are the aliphatic monocarboxylic and malonic acids and are more soluble at neutrality.

493-47-0, Fumaric acid, (1-hydroxytetradecyl)methyl-, 1-, γ-lactone

(bacteriostatic action of)

493-47-0 CAPLUS RN

L3 ANSWER 46 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1939:54740 CAPLUS DOCUMENT NUMBER: 33:54740

ORIGINAL REFERENCE NO.: 33:7885h-i,7886a

TITLE: The effects of agaricic, abietic and lichesteric acids

AUTHOR(S): Fischer, R.; Toth, D.

SOURCE: Archiv fuer Experimentelle Pathologie und

Pharmakologie (1938), 190, 500-9

CODEN: AEXPBL; ISSN: 0365-2041

DOCUMENT TYPE: Journal LANGUAGE: Unavailable

AB The hemolytic indexes for agaricic (I), lichesteric (II) and abietic (III)

acids were, resp.: 30,000, 40,000 and 18,000. On addition of cholesterol the hemolytic indexes for I, II and III were 1800, 5000 and 16,000. The foam values for I, II and III were 1:30,000, 1:25,000 and 1:1000. The absorption-increasing doses in  $\gamma$  per g. of frog for I, II and III were, resp.:  $5 \gamma$  after 55 min.,  $3 \gamma$  after 45 min. and 120

y after 150 min. The fish indexes were 1:25,000, 1:25,000 and

1:5000. 493-47-0P, Lichesteric acid

RL: PREP (Preparation)
(preparation of)

ΙT

RN 493-47-0 CAPLUS CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX

L3 ANSWER 47 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1937:21713 CAPLUS

DOCUMENT NUMBER: 31:21713

ORIGINAL REFERENCE NO.: 31:3028h-i,3029a-i

TITLE: Lichen substances. LXXVII. The lichen aliphatic acids from Nephromopsis endocrocea AUTHOR(S): Asahina, Yasuhiko; Yanaqita, Masaiti; Sakurai, Y.

Berichte der Deutschen Chemischen Gesellschaft [Abteilung] B: Abhandlungen (1937), 70B, 227-35

CODEN: BDCBAD; ISSN: 0365-9488
COCUMENT TYPE: Journal

DOCUMENT TYPE:

LANGUAGE: Unavailable

GI For diagram(s), see printed CA Issue.

It had been shown (C. A. 29, 7308.5) that Nephromopsis endocrocea Y. Asahina yields, in addition to the yellow pigment endocrocin, a colorless aliphatic acid (I) and a neutral substance (II). I, which was apparently a homogeneous lactonic acid, m. 93-5°, [a]D20 25.46°, proved to be really a mix. of 2 acids, for with KMnO4 it gave lauric acid and a saturated monobasic lactonic acid C17H30O4, designated nephrosteranic acid (III), and on ozonolysis yielded a considerable amount of HCHO, indicating the presence of a vinvl group (Clemo and MacDonald, C. A. 29, 7939.2). If I is heated with Ac20, it gives an acid (IV), m. 112°, [α]D24 33.75° (CHCl3), stable toward cold KMnO4 but partly oxidized to lauric acid when heated, leaving III. With boiling alkali IV partially changes into a ketonic acid, nephrosterylic acid, C16H30O3 (V), whose oily oxime gives on Beckmann rearrangement an amide which can be cleaved to undecylamine, m. 20° (Bz derivative, m. 57°), and pyrotartaric acid, m. 112°. On dry distillation IV gives, along with III, an unsatd. lactone, C16H28O2 (VI), which is hydrolyzed by alkali to V; it must therefore be the enol lactone of V and is called nephrosterylolactone. These facts show that III is an original component of I which remains unchanged in all the above reactions. The other (unsatd.) component, which is designated nephrosterinic acid (VII), is reminiscent of protolichesterinic acid (C. A. 26, 5067). To sep. III and VII, I was treated with semicarbazide, which gave, together with III, a semicarbazino compound, C18H33O5N3 (VIII); the free VII could not be regenerated from VIII, but on the assumption that the semicarbazide adds at the vinyl double bond, VII would have the composition C17H28O4. VII was also obtained as a Hq(OH) Cl compound (IX) by treating I with Hq(OAc)2 and then with NaCl; demercurization of IX yielded no well defined product, however. A sharp separation of III and VII was effected by chromatography on Al303, the unsatd. VII being retained in the upper part of the Al203 while III accumulated in the lower part. On catalytic hydrogenation, the mixture I was completely converted into III; III is therefore a dihydro derivative of VII. VII is accordingly assigned the structure shown in the accompanying formula. By rearrangement it changes into isonephrosterinic acid (X) which on distillation loses CO2 and gives VI. On saponification with

alkali,
both X and VI yield V, Cl1H23COCH2CHMeCO2H, whose structure was
established by synthesis as well as by the Hofmann rearrangement of its
oxime (see above). II is very similar to, perhaps identical with caperin
(J. prakt. Chemical 58, 409(1898)); it gives sterol-like color reactions, a
property which has not been reported for caperin. III (0.3 g. from 1 g. I
in 10% KOH treated with saturated KMnO4 to a permanent violet color), m.
95°, is recovered unchanged when boiled 3 hrs. in 10% KOH and
acidified. V, m. 74°, soluble without color in Na2CO3; semicarbazone,
m. 117°. VI (2.5 g. from 5 g. IV heated at 200-10° under 15
mm. until the evolution of CO2 ceases and then distilled at 210-30°),
b3 185-9°, decolorizes KMnO4. VIII (0.4 g. from 1 g. I), sinters
around 150°, decomposes 183-4°, is quite stable to KMnO4 in
acetone. IX, m. 95°, very stable to HCl, gives in alc. AcOH HBS
with H2S but the filtrate yields only amorphous products. VII, m.

96°, [a]D10 10.81° (CHCl3), instantly decolorizes KMnO4 in acetone. X (0.05 g. from 0.12 g. VII heated 1 hr. in Ac20 at 105°), m. 113°, [α]D11 32.98° (CHC13), stable to KMnO4 in acetone. Et laurinoylacetate (XI), from Et laurinoylacetoacetate and NH4OH, bl0 173-5° gives with PhNHNH2 phenylundecylpyrazolone, sandy powder becoming discolored at 205° and carbonizing around 240°. Heated 4 hrs. in alc. at 120° with Na and MeCHBrCO2Me, XI yields a light yellow oil, b4 180-90°, consisting chiefly of Me Et methyllaurinovlsuccinate, which, heated 8 hrs. with HI (d. 1.7) on the water bath, gives  $\alpha$ -methyl-β-laurinoylpropionic acid (= V). II, (C12H20O3)n, m. 248°, [α]D18.5 -100.2° (CHC13), insol. in KOH, gives no color in alc. with either FeCl3 or bleaching powder, dissolves in hot concentrated H2SO4 with red-brown color changing to dirty green; the CHC13 solution with a few drops Ac2O and 1 drop concentrated H2SO4 becomes blue-violet, then green. 75232-40-5P, Isonephrosterinic acid RL: PREP (Preparation) (preparation of)

3-Furancarboxvlic acid, 2,5-dihvdro-4-methvl-5-oxo-2-undecvl- (9CI) (CA

INDEX NAME)

75232-40-5 CAPLUS

RN

CN

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L3 ANSWER 48 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER:
                        1936:22403 CAPLUS
DOCUMENT NUMBER:
                         30:22403
ORIGINAL REFERENCE NO.: 30:2945i,2946a-q
TITLE:
                        Lichen substances, LXII, Constituents of Cetraria
                         islandica Ach.
AUTHOR(S):
                        Asahina, Yasuhiko; Yanagita, Masaiti
                        Berichte der Deutschen Chemischen Gesellschaft
                        [Abteilung] B: Abhandlungen (1936), 69B, 120-5
                        CODEN: BDCBAD: ISSN: 0365-9488
DOCUMENT TYPE:
                        Journal
LANGUAGE:
                        Unavailable
AB
    cf. C. A. 30, 1041.1. Asano (C. A. 26, 5067) established the structures
     of protolichesterinic (I) and lichesterinic acid (II), but as he worked
     not with Cetraria islandica Ach. (III) but with a lichen now considered to
     be an independent species, C. tenuifolia (Retz.) Howe (IV), the authors
     undertook a study of the true III, gathered on Mt. Asibetu and
     morphologically identical in all respects with the European lichen. It
     contained about 4% of a fatty acid mixture, m. around 90°,
     [α]D20 -45.62° (CHCl3), from which d-I was readily isolated.
     The mother liquor then vielded a strongly 1-rotatory isomer,
     1-alloprotolichesterinic acid (V), which gave 1-II with hot Ac20 and a
     pyrazoline derivative with CH2N2, and hence must be structurally identical
     with I. Heating the fatty acid mixture with Ac20 gave, as expected, d1-II.
     IV yielded 1-I. The fumaroprotocetraric acid, however, which is always
     found in the European III and in IV, could not be detected in the Japanese
     III. Theoretically, I has 4 possible different configurations (2 pairs of
     optical antipodes). There is no reason for assuming a change in the
     configuration at C atom 4 when I changes into II; 1-I would then differ
     from 1-V only in the configuration at C atom 3. Hydrogenation of the I
     gives, theoretically, 2 dihydro derivs. each, the 8 isomers forming 4
     pairs of optical antipodes. Whether the dihydro derivs. obtained from
     1-I, d-I and 1-V are homogeneous or mixts. of 2 diastereomers has not yet
     been established. d-I, m. 106°, [α]D20 12.07°
     (CHCl3). V, m. 88°, [\alpha]D23 - 56.34° (absolute alc.),
     [\alpha]D20 -49.53° (CHCl3), instantly decolorizes KMnO4 in
     acetone. Compound, C21H36O4N2, from V and CH2N2, m. 68-9°,
     [α]D18 -73.69°, stable toward KMnO4 in acetone. 1-II, m.
     123°, [α]D20 -25.06° (CHCl3). Dihydro derivative of 1-V,
    m. 92-3°, stable toward KMnO4, [α]D20 -7.41° (CHC13).
     1-I, m. 106°, [α]D18 -12.12° (CHCl3); dihydro derivative,
     m. 106°, [α]D18 -30.96° (CHCl3); pyrazoline derivative, m.
     54-5°, [α]D18 -183.1° (CHCl3). Dihydro derivative of d-I,
     m. 106°, [α]D15 34.60° (CHCl3); pyrazoline derivative, m.
     54-5°, [α]D18 190.60°.
     22800-25-5P, Lichesterinic acid, 1-
     RL: PREP (Preparation)
        (preparation of)
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3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2S)- (CA

INDEX NAME)
Absolute stereochemistry.

22800-25-5 CAPLUS

RN

CN

L3 ANSWER 49 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1935:39201 CAPLUS DOCUMENT NUMBER: 29:39201

ORIGINAL REFERENCE NO.: 29:5072d-f

TITLE: Constituents of Iceland moss. V. Reduction of

di-hydroprotolichesterinic acid and lichesterinic acid

AUTHOR(S): Asano, Michizo; Azumi, Tiaki

SOURCE: Berichte der Deutschen Chemischen Gesellschaft

[Abteilung] B: Abhandlungen (1935), 68B, 991-4

CODEN: BDCBAD; ISSN: 0365-9488

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

AB Cf. C. A. 26, 5067. λ-Isostearic acid (I), from lichesterinic acid with HI and red P (Boehm, Arch. Pharm. 241, 1 (1903)), m. 48-9°;

amide, m.  $104-4.5^{\circ}$ ; anilide, m.  $86-6.5^{\circ}$ ; p-toluide, m.  $82-3^{\circ}$ . Lichesterylic acid with N2H4.H2O gives

4-methyl-6-tridecylpyridazinone, m. 66°, which with NaOEt at

170-80° smoothly yields I. I was also synthesized by condensing

MeCH(CO2Et)2 with NaOEt and pentadecyl iodide to di-Et

methylpentadecylmalonate, yellowish oil, b2  $197-207^\circ$ , saponifying the ester to the free acid, m.  $95.5-6.5^\circ$ , decomposing about  $175^\circ$ ,

and decarboxylating the latter at 170-80°. There can be no doubt, therefore, that I is  $\alpha\text{-methylheptadecanoic acid.}$ 

Dihydro-d-protolichesterinic acid, m. 104-6° (Me ester, m. 51.5-2.5°), heated with HI and red P in a sealed tube and then reduced with Zn and AcOH, gives a-methyl-a'-tetradecylsuccinic

acid, m. 133-5°. 493-47-0, Lichesterinic acid

IT 493-47-0, Lichest (reduction of)

RN 493-47-0 CAPLUS

L3 ANSWER 50 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1932:49136 CAPLUS DOCUMENT NUMBER: 26:49136

ORIGINAL REFERENCE NO.: 26:5067f-h

TITLE: Constitution of protolichesterinic acid and

lichesterinic acid

AUTHOR(S): Asano, M.; Kanematsu, T.

DURCE: Berichte der Deutschen Chemischen Gesellschaft
[Abteilung] B: Abhandlungen (1932), 65B, 1175-8

CODEN: BDCBAD: ISSN: 0365-9488

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

GI For diagram(s), see printed CA Issue.

AB cf. C. A. 25, 4266-7. The reactions of protolichesterinic and

lichesterinic acid are best explained by the formulas I and II, (R = Me(CH2)12), resp., for the 2 acids. The following exptl. data are given in the present paper: II, m. 123.5 $^\circ$ , was obtained in 59-g. yield

in the present paper: 11, m. 123.5°, was obtained in 59-g. yield from 3800 g. Iceland moss from Tateyama, Province of Etchu. With excess

of 0.1 N KOH on the water bath it gives lichesterylic acid, m. 83-4° (semicarbazone, m. 125°). From 3 g. 1-I, m.

107.5°, with CH2N2 is obtained a neutral compound (III) m.

60-1°, which does not decolorize KMnO4, while 1-II forms only the

Me ester, C20H34O4, m.  $53-4^{\circ}$ , [ $\alpha$ ]D14  $-28.07^{\circ}$  (CHC13). II is strikingly stable toward KMnO4, but after long-continued action in

the cold it is finally converted into myristic acid.

IT 493-47-0P, Lichesterinic acid RL: PREP (Preparation)

(preparation of)

RN 493-47-0 CAPLUS

L3 ANSWER 51 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 1931:37818 CAPLUS DOCUMENT NUMBER: 25:37818 ORIGINAL REFERENCE NO.: 25:4266i,4267a-c TITLE: Constituents of Icelandic moss, III. Synthesis of

lichesteric acid

AUTHOR(S): Asano, M.; Ohta, Z.

Yakugaku Zasshi (1931), 51, 395-401(in German 36-7)

CODEN: YKKZAJ; ISSN: 0031-6903 Journal

DOCUMENT TYPE: LANGUAGE: Unavailable

The present work was undertaken to study the structure of protolichesteric acid (I). It had been shown that I on boiling with anhydrous AcOH gave lichesteric acid which on hydrolysis with alkali gave lichesterylic acid, C19H34O3, a keto acid. The oxime of this latter acid on Beckmann rearrangement gave an acid amide (II) which on hydrolysis gave N-tridecylamine and methylsuccinic acid. An attempt was made to determine the position of the Me group in II by synthesis. Myristyl chloride (prepared by treating myristic acid (20 g.) with SO2C12 (32 g.) when treated with NH3 in the cold gave the amide (III), m. 105-6° (yield 16 g.). III (16 g.) in MeOH (100 g.) when treated with NaOEt gave tridecylurethan (IV), C13H27NHCO2Me, m. 56° (vield 6 g.), which was hydrolyzed to tridecylamine (V). V (10 g.) in Et20 when treated with CH2ClCOCl (16.6 g.) for 1 hr. on the water bath gave chloroacetyltridecylamine (VI), C16H30ONCl, m. 66.5-7° (yield 8 g.). VI (6 g.) when treated with CH2(CO2Me)2 at 120° for 8 hrs. gave a compound (yield 10 g.) m. 69-70°, whose composition corresponded to C13H27NHCOCH2OC2H5. Myristyl chloride (30.5 g.) with AcCH2CO2Et (31 g.) and Na (5.4 g.) gave Et myristylacetoacetate (VII), b3 170-83° (yield 24.7 g.). This gave the characteristic  $\beta$ -ketone reactions. VII (8.5 g.) in absolute alc. (20 cc.) and Na (0.66 g.) with MeCHBrCO2Et (4.2 g.) in a sealed tube at 120° for 4 hrs. gave a compound (VIII) (yield 8.5 g.). Saponification of VIII with alc. KOH gave a compound m. 83-4° which did not depress the

126°. 22800-25-5P, Lichesterinic acid, 1-RL: PREP (Preparation)

(preparation of) 22800-25-5 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2S)- (CA INDEX NAME)

m. p. of the natural lichesterylic acid. The semicarbazone m.

Absolute stereochemistry.

L3 ANSWER 52 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 1931:37817 CAPLUS DOCUMENT NUMBER: 25:37817

ORIGINAL REFERENCE NO.: 25:4266g-i

TITLE: Constituents of Icelandic moss. II

Asano, M.; Kanematsu, T. AUTHOR(S):

Yakugaku Zasshi (1931), 51, 390-5 (in German 35) SOURCE:

CODEN: YKKZAJ; ISSN: 0031-6903

DOCUMENT TYPE: Journal

LANGUAGE: Unavailable

For diagram(s), see printed CA Issue.

cf. C. A. 22, 4470. In a previous investigation A. isolated from

Icelandic moss of Nikko province 1-protolichesteric acid, C19H22O4, m. 107.5-8°, for which he suggested the structure HO2CCH-c:cH2

Me(CH2)12CH.O.CO or HO2CC:CMe Me(CH2)12CH.O.CO. Using the same method, A. and K. isolated from Icelandic moss of Tateyama province a compound (I), m. 121-2°, [α]D15 -32.06°, which did not depress the m.

p. of 1-lichesteric acid, C19H32O4, m. 124°, isolated from

Icelandic moss of Nikko. I with 10% NaOH on the water bath for 2 hrs. gave lichesteric acid, m.  $83-4^{\circ}$ , which did not depress the m. p. of

the lichesteric acid obtained from Icelandic moss of Nikko. A mixture of equal quantities of 1- and d-protolichesteric acid (m. 107°) obtained from the European Icelandic moss, m. 100-1, [a]D10

±0°. 493-47-0P

RL: SPN (Synthetic preparation); PRP (Properties); PREP (Preparation) (Constituents of Icelandic moss. II)

RN 493-47-0 CAPLUS

CN 3-Furancarboxvlic acid, 2,5-dihvdro-4-methvl-5-oxo-2-tridecvl- (CA INDEX NAME)

22800-25-5P, Lichesterinic acid, 1-RL: PREP (Preparation)

(preparation of) RN 22800-25-5 CAPLUS

CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2S)- (CA INDEX NAME)

Absolute stereochemistry.

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L3 ANSWER 53 OF 53 CAPLUS COPYRIGHT 2009 ACS on STN
ACCESSION NUMBER:
                         1928:37595 CAPLUS
DOCUMENT NUMBER:
                         22:37595
ORIGINAL REFERENCE NO.: 22:4470g-i,4471a-c
TITLE:
                        Constitution of protolichestearic acid. I
                         Asahina, Y.; Asano, M.
AUTHOR(S):
CORPORATE SOURCE:
                        Tokyo Imp. Univ.
SOURCE:
                         Yakugaku Zasshi (1927), No. 539, 1-17
                         CODEN: YKKZAJ; ISSN: 0031-6903
DOCUMENT TYPE:
                         Journal
                         Unavailable
LANGUAGE:
GI
     For diagram(s), see printed CA Issue.
AB
     By Et20 extraction of Cetraria islandica Ach. f. anguslifolia, Kraplh., a
     subalpine moss in Japan, 1-protolichestearic acid (I), C19H32O4, m.
     105°, [α]D27 -12.71°, was isolated in 1.3% yield. It
     is the optical antipode of the d-acid found in European lichens. I, H2
     and Pt black gave dihydroprotolicheslearic acid, C19H34O4, m. 101°.
     I and H2NCONHNH2 gave the semicarbazone, m. about 140°. These
     reactions indicate the presence of a double bond in
     α,β-position to the CO group. Oxidation of I with KMnO4 gave
     myristic acid, while the oxidation with O3 and subsequent decomposition with
     H2O gave besides HCO2H and (CO2H)2, α-hydroxypentadecylic acid,
     C14H28(OH)CO2H. Heating of I with Ac2O resulted in an isometic change and
     gave 1-lichestearic acid (II), C19H32O4, m. 124°, [α]D25
     -32.66°. Heating of II with 10% KOH gave with CO2 evolution,
     lichesteryl acid (III), C18H34O3, m. 83-4°. III has previously
     been prepared by Sinnhold (Ann. 55, 144), but the nature of the third O atom
     remained unexplained. Heating of the oxime of III with H2SO4 resulted in
     Beckmann rearrangement and gave an acid amide (IV) C18H35(NO3), m.
     102°. IV and concentrated HBr in a closed tube gave tridecylamine and
     methylsuccinic acid. The above reactions show that III has 2 possible
     structures RCOCH2CHMeCO2H or RCOCHMeCH2CO2H(R = Me(CH2)12-). Heating of
     II in a vacuum at 20 mm. and 210° gave lichesteryl lactone (V), b.
     207°, which on saponification with KOH gave III. V, H2 and Pd-BaSO4 gave
     the dihydro derivative of V, m. 37-8°, while V, O3 and H2O gave AcOH as
     a decomposition product. Contrary to the view of Boehm (Arch. Pharm. 241, 1) V
     is therefore unsatd. The above reactions show that the relation of III to
     V is like that of levulinic acid to angelic lactone. Hence V has one of
     the following 4 possible structures: (a) R-CH.CH:CMe.CO.O, (b)
     R-C:CH.CHMe.CO.O, (c) RCH.CMe:CH.CO.O, (d) RC:C.Me.CH2.CO.O. But the fact
     that the ozonide of V gave AcOH instead of (CO2H)2 favors the structure
     (a) for V, while III should have the structure, RCOCH2CH(Me)CO2H. I,
     therefore, has one of the 2 possible structures, RCH.CH(CO2H).C(:CH2)CO.O
     or RCH.C(CO2H): CMe.CO.O. Since the ozonide of I gave HCO2H and (CO2H)2
     instead of AcOH, the former structure is preferred. From the fact that I
     did not give III, but II gave III by saponification with an alkali, the
following
     structure is assigned for III.
     493-47-0P
     RL: SPN (Synthetic preparation); PRP (Properties); RCT (Reactant); PREP
     (Preparation); RACT (Reactant or reagent)
        (Constitution of protolichestearic acid. I)
RN
     493-47-0 CAPLUS
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3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl- (CA INDEX

CN

NAME)

IT 22800-25-5P, Lichesterinic acid, 1-RL: PREP (Preparation)

- (preparation of) RN 22800-25-5 CAPLUS
- CN 3-Furancarboxylic acid, 2,5-dihydro-4-methyl-5-oxo-2-tridecyl-, (2S)- (CA INDEX NAME)

## Absolute stereochemistry.

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